

# Sebacic acid, butyl 2-hexyl ester

|                      |                                                                                   |
|----------------------|-----------------------------------------------------------------------------------|
| Inchi:               | InChI=1S/C20H38O4/c1-4-6-14-18(3)24-20(22)16-13-11-9-8-10-12-15-19(21)23-17-7-5-2 |
| InchiKey:            | HYSNYCBFLYZWAA-UHFFFAOYSA-N                                                       |
| Formula:             | C20H38O4                                                                          |
| SMILES:              | CCCCOC(=O)CCCCCCCCC(=O)OC(C)CCCC                                                  |
| Mol. weight [g/mol]: | 342.51                                                                            |

## Physical Properties

| Property code | Value   | Unit    | Source         |
|---------------|---------|---------|----------------|
| gf            | -352.76 | kJ/mol  | Joback Method  |
| hf            | -951.01 | kJ/mol  | Joback Method  |
| hfus          | 49.61   | kJ/mol  | Joback Method  |
| hvap          | 78.04   | kJ/mol  | Joback Method  |
| log10ws       | -6.03   |         | Crippen Method |
| logp          | 5.572   |         | Crippen Method |
| mcvol         | 307.540 | ml/mol  | McGowan Method |
| pc            | 1083.49 | kPa     | Joback Method  |
| rinpol        | 2299.00 |         | NIST Webbook   |
| rinpol        | 2299.00 |         | NIST Webbook   |
| tb            | 809.14  | K       | Joback Method  |
| tc            | 993.61  | K       | Joback Method  |
| tf            | 444.48  | K       | Joback Method  |
| vc            | 1.198   | m3/kmol | Joback Method  |

## Temperature Dependent Properties

| Property code | Value     | Unit    | Temperature [K] | Source        |
|---------------|-----------|---------|-----------------|---------------|
| cpg           | 969.66    | J/mol×K | 809.14          | Joback Method |
| cpg           | 987.99    | J/mol×K | 839.88          | Joback Method |
| cpg           | 1005.26   | J/mol×K | 870.63          | Joback Method |
| cpg           | 1021.49   | J/mol×K | 901.37          | Joback Method |
| cpg           | 1036.69   | J/mol×K | 932.12          | Joback Method |
| cpg           | 1050.89   | J/mol×K | 962.86          | Joback Method |
| cpg           | 1064.10   | J/mol×K | 993.61          | Joback Method |
| dvisc         | 0.0009916 | Pa×s    | 444.48          | Joback Method |

|       |           |      |        |               |
|-------|-----------|------|--------|---------------|
| dvisc | 0.0004441 | Pa×s | 505.26 | Joback Method |
| dvisc | 0.0002364 | Pa×s | 566.03 | Joback Method |
| dvisc | 0.0001422 | Pa×s | 626.81 | Joback Method |
| dvisc | 0.0000935 | Pa×s | 687.59 | Joback Method |
| dvisc | 0.0000659 | Pa×s | 748.36 | Joback Method |
| dvisc | 0.0000489 | Pa×s | 809.14 | Joback Method |

## Sources

|                        |                                                                                                                                           |
|------------------------|-------------------------------------------------------------------------------------------------------------------------------------------|
| <b>NIST Webbook:</b>   | <a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U355646&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U355646&amp;Units=SI</a> |
| <b>Crippen Method:</b> | <a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>                                 |
| <b>Crippen Method:</b> | <a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>                         |
| <b>Joback Method:</b>  | <a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>                                     |
| <b>McGowan Method:</b> | <a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>                     |

## Legend

|                 |                                                 |
|-----------------|-------------------------------------------------|
| <b>cpg:</b>     | Ideal gas heat capacity                         |
| <b>dvisc:</b>   | Dynamic viscosity                               |
| <b>gf:</b>      | Standard Gibbs free energy of formation         |
| <b>hf:</b>      | Enthalpy of formation at standard conditions    |
| <b>hfus:</b>    | Enthalpy of fusion at standard conditions       |
| <b>hvap:</b>    | Enthalpy of vaporization at standard conditions |
| <b>log10ws:</b> | Log10 of Water solubility in mol/l              |
| <b>logp:</b>    | Octanol/Water partition coefficient             |
| <b>mcvol:</b>   | McGowan's characteristic volume                 |
| <b>pc:</b>      | Critical Pressure                               |
| <b>rinpol:</b>  | Non-polar retention indices                     |
| <b>tb:</b>      | Normal Boiling Point Temperature                |
| <b>tc:</b>      | Critical Temperature                            |
| <b>tf:</b>      | Normal melting (fusion) point                   |
| <b>vc:</b>      | Critical Volume                                 |

Latest version available from:

<https://www.chemeo.com/cid/57-512-0/Sebacic-acid-butyl-2-hexyl-ester.pdf>

Generated by Cheméo on 2025-12-21 04:50:01.329518609 +0000 UTC m=+6040798.859559274.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.