

# (Dinitromethyl)benzene

**Inchi:** InChI=1S/C7H6N2O4/c10-8(11)7(9(12)13)6-4-2-1-3-5-6/h1-5,7H  
**InchiKey:** VMMLSJNPVNTYMN-UHFFFAOYSA-N  
**Formula:** C7H6N2O4  
**SMILES:** O=[N+](O-)C(c1ccccc1)[N+](=O)[O-]  
**Mol. weight [g/mol]:** 182.13  
**CAS:** 611-38-1

## Physical Properties

Property code	Value	Unit	Source
chs	-3571.00 ± 5.90	kJ/mol	NIST Webbook
gf	189.13	kJ/mol	Joback Method
hf	35.00 ± 6.70	kJ/mol	NIST Webbook
hfs	-41.00 ± 5.90	kJ/mol	NIST Webbook
hfus	27.13	kJ/mol	Joback Method
hsub	76.10 ± 0.80	kJ/mol	NIST Webbook
hvap	66.25	kJ/mol	Joback Method
log10ws	-2.98		Crippen Method
logp	1.239		Crippen Method
mcvol	120.570	ml/mol	McGowan Method
pc	4288.66	kPa	Joback Method
tb	689.48	K	Joback Method
tc	968.07	K	Joback Method
tf	467.29	K	Joback Method
vc	0.477	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	303.93	J/molxK	689.48	Joback Method
cpg	314.21	J/molxK	735.91	Joback Method
cpg	323.43	J/molxK	782.34	Joback Method
cpg	331.68	J/molxK	828.78	Joback Method
cpg	339.02	J/molxK	875.21	Joback Method
cpg	345.55	J/molxK	921.64	Joback Method

cpg	351.34	J/mol×K	968.07	Joback Method
hsubt	76.10	kJ/mol	317.50	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C611381&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C611381&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hsub:</b>	Enthalpy of sublimation at standard conditions
<b>hsubt:</b>	Enthalpy of sublimation at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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