

# Phenylacetic acid, 3-chloroprop-2-enyl ester

<b>Inchi:</b>	InChI=1S/C11H11ClO2/c12-7-4-8-14-11(13)9-10-5-2-1-3-6-10/h1-7H,8-9H2/b7-4+
<b>InchiKey:</b>	PBULRLQIJUOCDV-QPJXVBHSA-N
<b>Formula:</b>	C11H11ClO2
<b>SMILES:</b>	O=C(Cc1ccccc1)OCC=CCl
<b>Mol. weight [g/mol]:</b>	210.66

## Physical Properties

Property code	Value	Unit	Source
gf	-11.48	kJ/mol	Joback Method
hf	-177.16	kJ/mol	Joback Method
hfus	25.47	kJ/mol	Joback Method
hvap	55.86	kJ/mol	Joback Method
log10ws	-2.89		Crippen Method
logp	2.525		Crippen Method
mcvol	157.470	ml/mol	McGowan Method
pc	2823.32	kPa	Joback Method
rinpola	1563.20		NIST Webbook
tb	595.64	K	Joback Method
tc	818.39	K	Joback Method
tf	337.15	K	Joback Method
vc	0.597	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	353.13	J/molxK	595.64	Joback Method
cpg	409.48	J/molxK	781.26	Joback Method
cpg	399.80	J/molxK	744.14	Joback Method
cpg	389.37	J/molxK	707.01	Joback Method
cpg	378.14	J/molxK	669.89	Joback Method
cpg	366.08	J/molxK	632.76	Joback Method
cpg	418.45	J/molxK	818.39	Joback Method
dvisc	0.0001667	Paxs	595.64	Joback Method
dvisc	0.0002136	Paxs	552.56	Joback Method

dvisc	0.0002855	Paxs	509.48	Joback Method
dvisc	0.0004026	Paxs	466.39	Joback Method
dvisc	0.0006088	Paxs	423.31	Joback Method
dvisc	0.0010111	Paxs	380.23	Joback Method
dvisc	0.0019117	Paxs	337.15	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U292539&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U292539&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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