

# Benzene, 2,4-dibromo-1,3,5-trimethyl-

<b>Other names:</b>	2,4-Dibromomesitylene 2,4-dibromo-1,3,5-trimethylbenzene Mesitylene, 2,4-dibromo-
<b>Inchi:</b>	InChI=1S/C9H10Br2/c1-5-4-6(2)9(11)7(3)8(5)10/h4H,1-3H3
<b>InchiKey:</b>	CIHJFEWFZJQTFE-UHFFFAOYSA-N
<b>Formula:</b>	C9H10Br2
<b>SMILES:</b>	Cc1cc(C)c(Br)c(C)c1Br
<b>Mol. weight [g/mol]:</b>	277.98
<b>CAS:</b>	6942-99-0

## Physical Properties

Property code	Value	Unit	Source
gf	127.43	kJ/mol	Joback Method
hf	14.22	kJ/mol	Joback Method
hfus	22.12	kJ/mol	Joback Method
hvap	53.42	kJ/mol	Joback Method
log10ws	-5.22		Crippen Method
logp	4.137		Crippen Method
mcvol	148.910	ml/mol	McGowan Method
pc	3572.80	kPa	Joback Method
tb	551.70	K	NIST Webbook
tc	828.34	K	Joback Method
tf	387.29	K	Joback Method
vc	0.555	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	286.74	J/molxK	584.24	Joback Method
cpg	297.64	J/molxK	624.92	Joback Method
cpg	307.86	J/molxK	665.61	Joback Method
cpg	317.44	J/molxK	706.29	Joback Method
cpg	326.43	J/molxK	746.97	Joback Method
cpg	334.85	J/molxK	787.65	Joback Method

cpg	342.74	J/molxK	828.34	Joback Method
dvisc	0.0009758	Paxs	387.29	Joback Method
dvisc	0.0007129	Paxs	420.12	Joback Method
dvisc	0.0005451	Paxs	452.94	Joback Method
dvisc	0.0004322	Paxs	485.76	Joback Method
dvisc	0.0003529	Paxs	518.59	Joback Method
dvisc	0.0002952	Paxs	551.41	Joback Method
dvisc	0.0002519	Paxs	584.24	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.45085e+01
Coeff. B	-4.50381e+03
Coeff. C	-9.63160e+01
Temperature range (K), min.	413.02
Temperature range (K), max.	586.02

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C6942990&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C6942990&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions

<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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