

# methyl m-methoxy carbonylcinnamate

<b>Inchi:</b>	InChI=1S/C12H12O4/c1-15-11(13)7-6-9-4-3-5-10(8-9)12(14)16-2/h3-8H,1-2H3/b7-6+
<b>InchiKey:</b>	XPSVBFWYSAVLGD-VOTSOKGWSA-N
<b>Formula:</b>	C12H12O4
<b>SMILES:</b>	<chem>COC(=O)C=Cc1cccc(C(=O)OC)c1</chem>
<b>Mol. weight [g/mol]:</b>	220.22

## Physical Properties

Property code	Value	Unit	Source
gf	-234.68	kJ/mol	Joback Method
hf	-438.33	kJ/mol	Joback Method
hfus	26.26	kJ/mol	Joback Method
hvap	63.51	kJ/mol	Joback Method
log10ws	-2.24		Crippen Method
logp	1.659		Crippen Method
mcvol	166.760	ml/mol	McGowan Method
pc	2741.15	kPa	Joback Method
rinsol	1786.00		NIST Webbook
tb	662.36	K	Joback Method
tc	882.56	K	Joback Method
tf	403.18	K	Joback Method
vc	0.627	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	414.20	J/molxK	662.36	Joback Method
cpg	469.67	J/molxK	845.86	Joback Method
cpg	460.18	J/molxK	809.16	Joback Method
cpg	449.90	J/molxK	772.46	Joback Method
cpg	438.82	J/molxK	735.76	Joback Method
cpg	426.93	J/molxK	699.06	Joback Method
cpg	478.39	J/molxK	882.56	Joback Method
dvisc	0.0001282	Paxs	662.36	Joback Method
dvisc	0.0001606	Paxs	619.16	Joback Method

dvisc	0.0002080	Paxs	575.97	Joback Method
dvisc	0.0002809	Paxs	532.77	Joback Method
dvisc	0.0004001	Paxs	489.57	Joback Method
dvisc	0.0006102	Paxs	446.38	Joback Method
dvisc	0.0010188	Paxs	403.18	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R504999&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R504999&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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