

# 1-Hexanol, 4-methyl-

<b>Other names:</b>	4-Methyl-1-hexanol 4-Methylhexan-1-ol 4-Methylhexanol
<b>Inchi:</b>	InChI=1S/C7H16O/c1-3-7(2)5-4-6-8/h7-8H,3-6H2,1-2H3
<b>InchiKey:</b>	YNPVNLWKVZZBTM-UHFFFAOYSA-N
<b>Formula:</b>	C7H16O
<b>SMILES:</b>	CCC(C)CCCO
<b>Mol. weight [g/mol]:</b>	116.20
<b>CAS:</b>	818-49-5

## Physical Properties

Property code	Value	Unit	Source
gf	-131.20	kJ/mol	Joback Method
hf	-345.32	kJ/mol	Joback Method
hfus	14.45	kJ/mol	Joback Method
hvap	47.47	kJ/mol	Joback Method
log10ws	-1.77		Crippen Method
logp	1.805		Crippen Method
mcvol	115.360	ml/mol	McGowan Method
pc	3121.00	kPa	Joback Method
ripol	955.00		NIST Webbook
ripol	950.00		NIST Webbook
ripol	925.90		NIST Webbook
ripol	925.90		NIST Webbook
ripol	925.90		NIST Webbook
ripol	950.00		NIST Webbook
ripol	1434.00		NIST Webbook
ripol	1445.00		NIST Webbook
ripol	1443.00		NIST Webbook
ripol	1434.00		NIST Webbook
ripol	1414.00		NIST Webbook
ripol	1414.00		NIST Webbook
tb	446.15 ± 1.00	K	NIST Webbook
tb	436.15 ± 2.00	K	NIST Webbook
tc	615.06	K	Joback Method
tf	214.47	K	Joback Method
vc	0.441	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	307.23	J/molxK	615.06	Joback Method
cpg	247.92	J/molxK	451.30	Joback Method
cpg	258.79	J/molxK	478.59	Joback Method
cpg	269.25	J/molxK	505.89	Joback Method
cpg	279.32	J/molxK	533.18	Joback Method
cpg	289.00	J/molxK	560.47	Joback Method
cpg	298.30	J/molxK	587.77	Joback Method
dvisc	0.0002018	Paxs	451.30	Joback Method
dvisc	0.1594974	Paxs	214.47	Joback Method
dvisc	0.0221021	Paxs	253.94	Joback Method
dvisc	0.0052125	Paxs	293.41	Joback Method
dvisc	0.0017316	Paxs	332.88	Joback Method
dvisc	0.0007266	Paxs	372.36	Joback Method
dvisc	0.0003601	Paxs	411.83	Joback Method
hvapt	62.60	kJ/mol	398.00	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.51728e+01
Coeff. B	-3.55282e+03
Coeff. C	-1.09531e+02
Temperature range (K), min.	348.21
Temperature range (K), max.	469.81

## Sources

NIST Webbook:

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C818495&Units=SI>

The Yaws Handbook of Vapor Pressure:

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpolar:</b>	Non-polar retention indices
<b>ripolar:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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