

# cis-2-Methyl-butyl-1,3-dioxane

<b>Inchi:</b>	InChI=1S/C9H18O2/c1-3-4-5-9-6-10-8(2)11-7-9/h8-9H,3-7H2,1-2H3/t8-,9+
<b>InchiKey:</b>	IJCIOLXHDMCLLI-DTORHVGOSA-N
<b>Formula:</b>	C9H18O2
<b>SMILES:</b>	CCCCC1COC(C)OC1
<b>Mol. weight [g/mol]:</b>	158.24
<b>CAS:</b>	39087-21-3

## Physical Properties

Property code	Value	Unit	Source
chl	-5504.00 ± 12.00	kJ/mol	NIST Webbook
gf	-130.60	kJ/mol	Joback Method
hf	-459.11	kJ/mol	Joback Method
hfl	-610.00 ± 12.00	kJ/mol	NIST Webbook
hfus	27.93	kJ/mol	Joback Method
hvap	44.77	kJ/mol	Joback Method
log10ws	-2.02		Crippen Method
logp	2.186		Crippen Method
mcvol	138.550	ml/mol	McGowan Method
pc	2690.21	kPa	Joback Method
tb	474.10	K	Joback Method
tc	671.99	K	Joback Method
tf	247.47	K	Joback Method
vc	0.513	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	321.52	J/molxK	474.10	Joback Method
cpg	402.07	J/molxK	639.01	Joback Method
cpg	387.53	J/molxK	606.03	Joback Method
cpg	372.22	J/molxK	573.04	Joback Method
cpg	356.12	J/molxK	540.06	Joback Method
cpg	339.22	J/molxK	507.08	Joback Method
cpg	415.85	J/molxK	671.99	Joback Method

dvisc	0.0003243	Paxs	474.10	Joback Method
dvisc	0.0004255	Paxs	436.33	Joback Method
dvisc	0.0005879	Paxs	398.56	Joback Method
dvisc	0.0008692	Paxs	360.78	Joback Method
dvisc	0.0014079	Paxs	323.01	Joback Method
dvisc	0.0025914	Paxs	285.24	Joback Method
dvisc	0.0057462	Paxs	247.47	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C39087213&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C39087213&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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