

# Propanedioic acid, (acetylamino)-, diethyl ester

**Other names:**

(Acetylamino)malonic acid diethyl ester  
(Acetylamino)propanedioic acid diethyl ester  
2-(Acetylamino)propanedioic acid diethyl ester  
Acetamidomalonic acid diethyl ester  
Acetaminomalonic acid diethyl ester  
Diethyl 2-acetamidomalonate  
Diethyl N-acetylaminomalonate  
Diethyl a-(acetylamino)malonate  
Diethyl acetamidomalonate  
Diethyl acetaminomalonate  
Diethyl acetylaminomalonate  
Diethylester kyseliny acetylaminomalonove  
Ethyl acetamidomalonate  
Malonic acid, acetamido-, diethyl ester  
NSC 7645  
Propanedioic acid, 2-(acetylamino)-, 1,3-diethyl ester

**Inchi:**

InChI=1S/C9H15NO5/c1-4-14-8(12)7(10-6(3)11)9(13)15-5-2/h7H,4-5H2,1-3H3,(H,10,11)

**InchiKey:**

ISOLMABRZPQKOV-UHFFFAOYSA-N

**Formula:**

C9H15NO5

**SMILES:**

CCOC(=O)C(NC(C)=O)C(=O)OCC

**Mol. weight [g/mol]:**

217.22

**CAS:**

1068-90-2

## Physical Properties

Property code	Value	Unit	Source
gf	-484.91	kJ/mol	Joback Method
hf	-783.08	kJ/mol	Joback Method
hfus	27.81	kJ/mol	Joback Method
hvap	66.73	kJ/mol	Joback Method
log10ws	-0.39		Crippen Method
logp	-0.383		Crippen Method
mcvol	164.100	ml/mol	McGowan Method
pc	2752.67	kPa	Joback Method
tb	661.50	K	Joback Method
tc	855.57	K	Joback Method
tf	423.10	K	Joback Method

tt	368.72	K	Heat Capacity and Thermodynamic Property Studies of Diethyl Acetamidomalonate (C <sub>9</sub> H <sub>15</sub> NO <sub>5</sub> )
vc	0.623	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	434.44	J/mol×K	661.50	Joback Method
cpg	446.19	J/mol×K	693.85	Joback Method
cpg	457.31	J/mol×K	726.19	Joback Method
cpg	467.79	J/mol×K	758.54	Joback Method
cpg	477.63	J/mol×K	790.88	Joback Method
cpg	486.80	J/mol×K	823.23	Joback Method
cpg	495.32	J/mol×K	855.57	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	458.20	K	2.70	NIST Webbook

## Sources

McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1068902&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1068902&amp;Units=SI</a>
Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
Heat Capacity and Thermodynamic Property Studies of Diethyl Acetamidomalonate (C <sub>9</sub> H <sub>15</sub> NO <sub>5</sub> ):	<a href="https://www.doi.org/10.1021/je400159e">https://www.doi.org/10.1021/je400159e</a>
Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h vap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>tt:</b>	Triple Point Temperature
<b>vc:</b>	Critical Volume

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