

# Carbosulfan

<b>Other names:</b>	Carbamic acid, [(dibutylamino)thio]methyl-, 2,3-dihydro-2,2-dimethyl-7-benzofuranyl ester, 2,3-Dihydro-2,2-dimethyl-7-benzofuranyl[(dibutylamino)thio]methyl carbamate Advantage FMC 35001 Marshal OMS 3022 Posse Sheriff 2,3-dihydro-2,2-dimethyl-7-benzofuryl [(dibutylamino)thio]methylcarbamate
<b>Inchi:</b>	InChI=1S/C20H32N2O3S/c1-6-8-13-22(14-9-7-2)26-21(5)19(23)24-17-12-10-11-16-15-2
<b>InchiKey:</b>	JLQUFIHWVLZVTJ-UHFFFAOYSA-N
<b>Formula:</b>	C20H32N2O3S
<b>SMILES:</b>	CCCCN(CCCC)SN(C)C(=O)Oc1cccc2c1OC(C)(C)C2
<b>Mol. weight [g/mol]:</b>	380.55
<b>CAS:</b>	55285-14-8

## Physical Properties

Property code	Value	Unit	Source
gf	200.57	kJ/mol	Joback Method
hf	-354.37	kJ/mol	Joback Method
hfus	53.59	kJ/mol	Joback Method
hvap	87.05	kJ/mol	Joback Method
log10ws	-6.46		Crippen Method
logp	5.296		Crippen Method
mcvol	307.660	ml/mol	McGowan Method
pc	1465.73	kPa	Joback Method
rinpol	2407.00		NIST Webbook
rinpol	2407.00		NIST Webbook
tb	897.52	K	Joback Method
tc	1114.32	K	Joback Method
tf	606.53	K	Joback Method
vc	1.137	m3/kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	990.16	J/mol×K	897.52	Joback Method
cpg	1009.21	J/mol×K	933.65	Joback Method
cpg	1027.87	J/mol×K	969.79	Joback Method
cpg	1046.31	J/mol×K	1005.92	Joback Method
cpg	1064.69	J/mol×K	1042.05	Joback Method
cpg	1083.18	J/mol×K	1078.19	Joback Method
cpg	1101.94	J/mol×K	1114.32	Joback Method

## Sources

Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C55285148&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C55285148&amp;Units=SI</a>
Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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