

# Benzaldehyde, 4-(diethylamino)-

<b>Other names:</b>	Benzaldehyde, p-(diethylamino)- p-(Diethylamino)benzaldehyde p-Formyl-N,N-diethylaniline 4-(Diethylamino)benzaldehyde 4-(N,N-Diethylamino)benzaldehyde
<b>Inchi:</b>	InChI=1S/C11H15NO/c1-3-12(4-2)11-7-5-10(9-13)6-8-11/h5-9H,3-4H2,1-2H3
<b>InchiKey:</b>	MNFZZNNFORDXSV-UHFFFAOYSA-N
<b>Formula:</b>	C11H15NO
<b>SMILES:</b>	CCN(CC)c1ccc(C=O)cc1
<b>Mol. weight [g/mol]:</b>	177.24
<b>CAS:</b>	120-21-8

## Physical Properties

Property code	Value	Unit	Source
gf	155.78	kJ/mol	Joback Method
hf	-63.36	kJ/mol	Joback Method
hfus	23.21	kJ/mol	Joback Method
hvap	51.78	kJ/mol	Joback Method
log10ws	-2.46		Crippen Method
logp	2.345		Crippen Method
mcvol	153.640	ml/mol	McGowan Method
pc	2811.36	kPa	Joback Method
tb	543.84	K	Joback Method
tc	748.45	K	Joback Method
tf	314.00	K	NIST Webbook
vc	0.579	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	357.09	J/molxK	543.84	Joback Method
cpg	371.85	J/molxK	577.94	Joback Method
cpg	385.74	J/molxK	612.04	Joback Method
cpg	398.80	J/molxK	646.14	Joback Method

cpg	411.06	J/mol×K	680.24	Joback Method
cpg	422.56	J/mol×K	714.34	Joback Method
cpg	433.34	J/mol×K	748.45	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	447.20	K	0.90	NIST Webbook
tbrp	447.00	K	0.90	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C120218&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C120218&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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