

# Chalcone

<b>Other names:</b>	2-Propen-1-one, 1,3-diphenyl- «alpha»-Benzylideneacetophenone «beta»-Benzoylstyrene «beta»-Phenylacrylophenone Benzalacetophenone Benzylideneacetophenone Benzylidenacetophenone Chalkone Cinnamophenone Phenyl styryl ketone 1-Phenyl-2-benzoylethylene 1,3-Diphenyl-1-propen-3-one 2-Benzalacetophenone 2-Benzylideneacetophenone 3-Phenylacrylophenone 1,3-Diphenyl-2-propen-1-one Acrylophenone, 3-phenyl- 1-Benzoyl-2-phenylethene 1,3-Diphenylpropenone Phenyl 2-phenylvinyl ketone Styryl phenyl ketone 1-Benzoyl-2-phenylethylene «omega»-Benzylideneacetophenone 1,3-Diphenyl-2-propenone NSC 26612
<b>Inchi:</b>	InChI=1S/C15H12O/c16-15(14-9-5-2-6-10-14)12-11-13-7-3-1-4-8-13/h1-12H
<b>InchiKey:</b>	DQFBYFPFKXHELB-UHFFFAOYSA-N
<b>Formula:</b>	C15H12O
<b>SMILES:</b>	O=C(C=Cc1ccccc1)c1ccccc1
<b>Mol. weight [g/mol]:</b>	208.26
<b>CAS:</b>	94-41-7

## Physical Properties

Property code	Value	Unit	Source
chs	-7626.20	kJ/mol	NIST Webbook
gf	251.54	kJ/mol	Joback Method

hf	124.77		kJ/mol	Joback Method
hfs	-24.00		kJ/mol	NIST Webbook
hfus	24.49		kJ/mol	Joback Method
hvap	60.24		kJ/mol	Joback Method
log10ws	-4.19			Crippen Method
logp	3.583			Crippen Method
mcvol	171.960		ml/mol	McGowan Method
pc	2826.33		kPa	Joback Method
rinpol	1970.60			NIST Webbook
rinpol	1970.60			NIST Webbook
tb	653.99		K	Joback Method
tc	905.95		K	Joback Method
tf	329.00 ± 4.00		K	NIST Webbook
vc	0.645		m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	419.04	J/molxK	653.99	Joback Method
cpg	434.68	J/molxK	695.98	Joback Method
cpg	448.96	J/molxK	737.98	Joback Method
cpg	462.00	J/molxK	779.97	Joback Method
cpg	473.92	J/molxK	821.97	Joback Method
cpg	484.83	J/molxK	863.96	Joback Method
cpg	494.84	J/molxK	905.95	Joback Method
dvisc	0.0019321	Paxs	356.50	Joback Method
dvisc	0.0009699	Paxs	406.08	Joback Method
dvisc	0.0005657	Paxs	455.66	Joback Method
dvisc	0.0003668	Paxs	505.25	Joback Method
dvisc	0.0002569	Paxs	554.83	Joback Method
dvisc	0.0001908	Paxs	604.41	Joback Method
dvisc	0.0001483	Paxs	653.99	Joback Method

## Sources

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C94417&Units=SI>

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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