

# Menthenal isomer

<b>Inchi:</b>	InChI=1S/C10H16O/c1-8-3-5-10(6-4-8)9(2)7-11/h7,9-10H,1,3-6H2,2H3
<b>InchiKey:</b>	JNRATHVXQNNAX-UHFFFAOYSA-N
<b>Formula:</b>	C10H16O
<b>SMILES:</b>	<chem>C=C1CCC(C(C)C=O)CC1</chem>
<b>Mol. weight [g/mol]:</b>	152.23

## Physical Properties

Property code	Value	Unit	Source
gf	8.89	kJ/mol	Joback Method
hf	-202.03	kJ/mol	Joback Method
hfus	11.10	kJ/mol	Joback Method
hvap	44.77	kJ/mol	Joback Method
log10ws	-2.55		Crippen Method
logp	2.568		Crippen Method
mcvol	138.170	ml/mol	McGowan Method
pc	2853.57	kPa	Joback Method
rinpol	1098.00		NIST Webbook
rinpol	1110.00		NIST Webbook
rinpol	1098.00		NIST Webbook
tb	495.13	K	Joback Method
tc	703.60	K	Joback Method
tf	250.52	K	Joback Method
vc	0.523	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	311.64	J/molxK	495.13	Joback Method
cpg	328.62	J/molxK	529.88	Joback Method
cpg	344.71	J/molxK	564.62	Joback Method
cpg	359.93	J/molxK	599.37	Joback Method
cpg	374.31	J/molxK	634.11	Joback Method
cpg	387.86	J/molxK	668.86	Joback Method
cpg	400.60	J/molxK	703.60	Joback Method

dvisc	0.0056040	Paxs	250.52	Joback Method
dvisc	0.0024868	Paxs	291.29	Joback Method
dvisc	0.0013471	Paxs	332.06	Joback Method
dvisc	0.0008345	Paxs	372.82	Joback Method
dvisc	0.0005681	Paxs	413.59	Joback Method
dvisc	0.0004144	Paxs	454.36	Joback Method
dvisc	0.0003184	Paxs	495.13	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R224876&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R224876&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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