

# Carbonic acid, allyl 4-benzyloxyphenyl ester

**Inchi:** InChI=1S/C17H16O4/c1-2-12-19-17(18)21-16-10-8-15(9-11-16)20-13-14-6-4-3-5-7-14/h2  
**InchiKey:** GKDJASGAZGOOQZ-UHFFFAOYSA-N  
**Formula:** C17H16O4  
**SMILES:** C=CCOC(=O)Oc1ccc(OCc2ccccc2)cc1  
**Mol. weight [g/mol]:** 284.31

## Physical Properties

Property code	Value	Unit	Source
gf	-48.63	kJ/mol	Joback Method
hf	-316.43	kJ/mol	Joback Method
hfus	31.36	kJ/mol	Joback Method
hvap	71.96	kJ/mol	Joback Method
log10ws	-4.77		Crippen Method
logp	3.967		Crippen Method
mcvol	217.750	ml/mol	McGowan Method
pc	2165.35	kPa	Joback Method
rinpol	2307.00		NIST Webbook
tb	764.51	K	Joback Method
tc	992.34	K	Joback Method
tf	461.57	K	Joback Method
vc	0.812	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	600.21	J/molxK	764.51	Joback Method
cpg	614.91	J/molxK	802.48	Joback Method
cpg	628.40	J/molxK	840.45	Joback Method
cpg	640.70	J/molxK	878.43	Joback Method
cpg	651.83	J/molxK	916.40	Joback Method
cpg	661.80	J/molxK	954.37	Joback Method
cpg	670.64	J/molxK	992.34	Joback Method
dvisc	0.0005752	Paxs	461.57	Joback Method
dvisc	0.0003390	Paxs	512.06	Joback Method

dvisc	0.0002197	Paxs	562.55	Joback Method
dvisc	0.0001530	Paxs	613.04	Joback Method
dvisc	0.0001125	Paxs	663.53	Joback Method
dvisc	0.0000864	Paxs	714.02	Joback Method
dvisc	0.0000688	Paxs	764.51	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U357813&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U357813&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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