

# Thiocyclohexane, 4-propylidene

<b>Inchi:</b>	InChI=1S/C8H14S/c1-2-3-8-4-6-9-7-5-8/h3H,2,4-7H2,1H3
<b>InchiKey:</b>	HPNXVYNVRLPFBY-UHFFFAOYSA-N
<b>Formula:</b>	C8H14S
<b>SMILES:</b>	CCC=C1CCSCC1
<b>Mol. weight [g/mol]:</b>	142.26

## Physical Properties

Property code	Value	Unit	Source
gf	133.96	kJ/mol	Joback Method
hf	-12.50	kJ/mol	Joback Method
hfus	11.22	kJ/mol	Joback Method
hvap	40.74	kJ/mol	Joback Method
log10ws	-2.80		Crippen Method
logp	2.850		Crippen Method
mvol	124.770	ml/mol	McGowan Method
pc	3360.64	kPa	Joback Method
rmpol	1134.00		NIST Webbook
ripol	1385.00		NIST Webbook
tb	461.13	K	Joback Method
tc	686.10	K	Joback Method
tf	285.35	K	Joback Method
vc	0.447	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	247.77	J/mol×K	461.13	Joback Method
cpg	263.78	J/mol×K	498.63	Joback Method
cpg	278.83	J/mol×K	536.12	Joback Method
cpg	292.97	J/mol×K	573.62	Joback Method
cpg	306.24	J/mol×K	611.11	Joback Method
cpg	318.68	J/mol×K	648.61	Joback Method
cpg	330.34	J/mol×K	686.10	Joback Method

# Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R384488&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R384488&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>

# Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mc<sub>vol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>ri<sub>npol</sub>:</b>	Non-polar retention indices
<b>ri<sub>pol</sub>:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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