

# 1H-Cyclopropa[l]phenanthrene,1a,9b-dihydro-

<b>Inchi:</b>	InChI=1S/C15H12/c1-3-7-12-10(5-1)11-6-2-4-8-13(11)15-9-14(12)15/h1-8,14-15H,9H2
<b>InchiKey:</b>	HJUMUYNFEXGVRH-UHFFFAOYSA-N
<b>Formula:</b>	C15H12
<b>SMILES:</b>	<chem>c1ccc2c(c1)-c1ccccc1C1CC21</chem>
<b>Mol. weight [g/mol]:</b>	192.26
<b>CAS:</b>	949-41-7

## Physical Properties

Property code	Value	Unit	Source
gf	438.78	kJ/mol	Joback Method
hf	261.27	kJ/mol	Joback Method
hfus	24.48	kJ/mol	Joback Method
hvap	54.17	kJ/mol	Joback Method
ie	7.77	eV	NIST Webbook
log10ws	-4.98		Crippen Method
logp	3.938		Crippen Method
mvol	152.970	ml/mol	McGowan Method
pc	2950.48	kPa	Joback Method
tb	606.59	K	Joback Method
tc	853.17	K	Joback Method
tf	383.13	K	Joback Method
vc	0.598	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	387.83	J/molxK	606.59	Joback Method
cpg	456.15	J/molxK	812.07	Joback Method
cpg	444.51	J/molxK	770.97	Joback Method
cpg	432.05	J/molxK	729.88	Joback Method
cpg	418.57	J/molxK	688.78	Joback Method
cpg	403.90	J/molxK	647.69	Joback Method
cpg	467.16	J/molxK	853.17	Joback Method
dvisc	0.0020667	Paxs	606.59	Joback Method

dvisc	0.0020701	Paxs	569.35	Joback Method
dvisc	0.0020740	Paxs	532.10	Joback Method
dvisc	0.0020785	Paxs	494.86	Joback Method
dvisc	0.0020838	Paxs	457.62	Joback Method
dvisc	0.0020900	Paxs	420.37	Joback Method
dvisc	0.0020975	Paxs	383.13	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C949417&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C949417&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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