

# Glycerol, 2,3-dimethyl, 1-(12-methyltetradecanoate)

Inchi:	InChI=1S/C20H40O4/c1-5-18(2)14-12-10-8-6-7-9-11-13-15-20(21)24-17-19(23-4)16-22-3
InchiKey:	BGVGITAXRXEPPX-UHFFFAOYSA-N
Formula:	C20H40O4
SMILES:	CCC(C)CCCCCCCCCCC(=O)OCC(COC)OC
Mol. weight [g/mol]:	344.53

## Physical Properties

Property code	Value	Unit	Source
gf	-331.28	kJ/mol	Joback Method
hf	-975.93	kJ/mol	Joback Method
hfus	45.67	kJ/mol	Joback Method
hvap	73.31	kJ/mol	Joback Method
log10ws	-5.10		Crippen Method
logp	5.138		Crippen Method
mcvol	311.840	ml/mol	McGowan Method
pc	1033.90	kPa	Joback Method
rinsol	2143.00		NIST Webbook
tb	777.25	K	Joback Method
tc	955.67	K	Joback Method
tf	401.78	K	Joback Method
vc	1.204	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	975.32	J/molxK	777.25	Joback Method
cpg	994.88	J/molxK	806.99	Joback Method
cpg	1013.39	J/molxK	836.72	Joback Method
cpg	1030.87	J/molxK	866.46	Joback Method
cpg	1047.31	J/molxK	896.20	Joback Method
cpg	1062.73	J/molxK	925.94	Joback Method
cpg	1077.13	J/molxK	955.67	Joback Method
dvisc	0.0011313	Paxs	401.78	Joback Method
dvisc	0.0004303	Paxs	464.36	Joback Method

dvisc	0.0002059	Paxs	526.94	Joback Method
dvisc	0.0001152	Paxs	589.51	Joback Method
dvisc	0.0000721	Paxs	652.09	Joback Method
dvisc	0.0000490	Paxs	714.67	Joback Method
dvisc	0.0000354	Paxs	777.25	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R56393&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R56393&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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