

# Succinic acid, 3,5-dichlorophenyl isobutyl ester

<b>Inchi:</b>	InChI=1S/C14H16Cl2O4/c1-9(2)8-19-13(17)3-4-14(18)20-12-6-10(15)5-11(16)7-12/h5-7,
<b>InchiKey:</b>	KJLLZEAVRZFCY-UHFFFAOYSA-N
<b>Formula:</b>	C14H16Cl2O4
<b>SMILES:</b>	CC(C)COC(=O)CCC(=O)Oc1cc(Cl)cc(Cl)c1
<b>Mol. weight [g/mol]:</b>	319.18

## Physical Properties

Property code	Value	Unit	Source
gf	-333.99	kJ/mol	Joback Method
hf	-645.06	kJ/mol	Joback Method
hfus	35.72	kJ/mol	Joback Method
hvap	77.05	kJ/mol	Joback Method
log10ws	-4.29		Crippen Method
logp	3.878		Crippen Method
mvol	223.720	ml/mol	McGowan Method
pc	2001.91	kPa	Joback Method
rinpol	2106.00		NIST Webbook
tb	783.36	K	Joback Method
tc	1000.10	K	Joback Method
tf	488.16	K	Joback Method
vc	0.852	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	589.53	J/molxK	783.36	Joback Method
cpg	601.85	J/molxK	819.48	Joback Method
cpg	613.20	J/molxK	855.61	Joback Method
cpg	623.60	J/molxK	891.73	Joback Method
cpg	633.05	J/molxK	927.85	Joback Method
cpg	641.56	J/molxK	963.98	Joback Method
cpg	649.13	J/molxK	1000.10	Joback Method
dvisc	0.0006953	Paxs	488.16	Joback Method
dvisc	0.0004209	Paxs	537.36	Joback Method

dvisc	0.0002771	Paxs	586.56	Joback Method
dvisc	0.0001947	Paxs	635.76	Joback Method
dvisc	0.0001439	Paxs	684.96	Joback Method
dvisc	0.0001107	Paxs	734.16	Joback Method
dvisc	0.0000881	Paxs	783.36	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U349720&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U349720&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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