

# Methyl 6-hydroxy-2-methyl-3,4-methylenedioxy-benzoate

<b>Inchi:</b>	InChI=1S/C10H10O5/c1-5-8(10(12)13-2)6(11)3-7-9(5)15-4-14-7/h3,11H,4H2,1-2H3
<b>InchiKey:</b>	VPYDVSWMSWPSKV-UHFFFAOYSA-N
<b>Formula:</b>	C10H10O5
<b>SMILES:</b>	<chem>COC(=O)c1c(O)cc2c(c1C)OCO2</chem>
<b>Mol. weight [g/mol]:</b>	210.18

## Physical Properties

Property code	Value	Unit	Source
gf	-375.48	kJ/mol	Joback Method
hf	-640.58	kJ/mol	Joback Method
hfus	36.12	kJ/mol	Joback Method
hvap	73.53	kJ/mol	Joback Method
log10ws	-1.95		Crippen Method
logp	1.216		Crippen Method
mcvol	142.190	ml/mol	McGowan Method
pc	4151.61	kPa	Joback Method
rinpol	1626.00		NIST Webbook
rinpol	1626.00		NIST Webbook
rinpol	1618.00		NIST Webbook
tb	692.04	K	Joback Method
tc	927.74	K	Joback Method
tf	525.64	K	Joback Method
vc	0.477	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	381.42	J/molxK	692.04	Joback Method
cpg	391.69	J/molxK	731.32	Joback Method
cpg	401.33	J/molxK	770.61	Joback Method
cpg	410.43	J/molxK	809.89	Joback Method
cpg	419.08	J/molxK	849.17	Joback Method
cpg	427.36	J/molxK	888.45	Joback Method
cpg	435.37	J/molxK	927.74	Joback Method

dvisc	0.0002744	Paxs	525.64	Joback Method
dvisc	0.0001839	Paxs	553.37	Joback Method
dvisc	0.0001280	Paxs	581.11	Joback Method
dvisc	0.0000921	Paxs	608.84	Joback Method
dvisc	0.0000682	Paxs	636.57	Joback Method
dvisc	0.0000518	Paxs	664.31	Joback Method
dvisc	0.0000402	Paxs	692.04	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R273902&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R273902&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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