

# Benzene, 1-cyclopropyl-4-methyl-

<b>Inchi:</b>	InChI=1S/C10H12/c1-8-2-4-9(5-3-8)10-6-7-10/h2-5,10H,6-7H2,1H3
<b>InchiKey:</b>	OHSOXYBIKPCSCK-UHFFFAOYSA-N
<b>Formula:</b>	C10H12
<b>SMILES:</b>	Cc1ccc(C2CC2)cc1
<b>Mol. weight [g/mol]:</b>	132.20
<b>CAS:</b>	6921-43-3

## Physical Properties

Property code	Value	Unit	Source
affp	846.30	kJ/mol	NIST Webbook
basg	813.80	kJ/mol	NIST Webbook
gf	196.85	kJ/mol	Joback Method
hf	48.13	kJ/mol	Joback Method
hfus	13.44	kJ/mol	Joback Method
hvap	40.70	kJ/mol	Joback Method
log10ws	-3.03		Crippen Method
logp	2.872		Crippen Method
mcvol	117.140	ml/mol	McGowan Method
pc	3333.53	kPa	Joback Method
tb	466.60	K	Joback Method
tc	689.90	K	Joback Method
tf	259.34	K	Joback Method
vc	0.445	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	242.54	J/mol×K	466.60	Joback Method
cpg	312.26	J/mol×K	652.68	Joback Method
cpg	300.22	J/mol×K	615.47	Joback Method
cpg	287.30	J/mol×K	578.25	Joback Method
cpg	273.42	J/mol×K	541.03	Joback Method
cpg	258.52	J/mol×K	503.82	Joback Method
cpg	323.47	J/mol×K	689.90	Joback Method

dvisc	0.0003992	Paxs	466.60	Joback Method
dvisc	0.0004515	Paxs	432.06	Joback Method
dvisc	0.0005217	Paxs	397.51	Joback Method
dvisc	0.0006195	Paxs	362.97	Joback Method
dvisc	0.0007629	Paxs	328.43	Joback Method
dvisc	0.0009866	Paxs	293.88	Joback Method
dvisc	0.0013663	Paxs	259.34	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C6921433&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C6921433&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>affp:</b>	Proton affinity
<b>basg:</b>	Gas basicity
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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