

# Cyclopentane, 1-ethyl-3-methyl-

<b>Other names:</b>	1-Ethyl-3-methylcyclopentane 1-Methyl-3-ethylcyclopentane
<b>Inchi:</b>	InChI=1S/C8H16/c1-3-8-5-4-7(2)6-8/h7-8H,3-6H2,1-2H3
<b>InchiKey:</b>	PQXAPVOKLYINEI-UHFFFAOYSA-N
<b>Formula:</b>	C8H16
<b>SMILES:</b>	CCC1CCC(C)C1
<b>Mol. weight [g/mol]:</b>	112.21
<b>CAS:</b>	3726-47-4

## Physical Properties

Property code	Value	Unit	Source
gf	45.32	kJ/mol	Joback Method
hf	-168.31	kJ/mol	Joback Method
hfus	11.48	kJ/mol	Joback Method
hvap	33.35	kJ/mol	Joback Method
log10ws	-2.58		Crippen Method
logp	2.833		Crippen Method
mcvol	112.720	ml/mol	McGowan Method
pc	2956.90	kPa	Joback Method
rinpol	785.00		NIST Webbook
rinpol	790.00		NIST Webbook
rinpol	793.90		NIST Webbook
rinpol	788.00		NIST Webbook
rinpol	808.00		NIST Webbook
rinpol	787.00		NIST Webbook
rinpol	787.00		NIST Webbook
rinpol	793.90		NIST Webbook
rinpol	787.00		NIST Webbook
rinpol	792.00		NIST Webbook
rinpol	785.00		NIST Webbook
tb	394.24 ± 0.70	K	NIST Webbook
tb	394.18 ± 0.20	K	NIST Webbook
tc	585.20	K	Joback Method
tf	145.13 ± 0.15	K	NIST Webbook
tf	140.40 ± 1.00	K	NIST Webbook
tf	140.35 ± 2.00	K	NIST Webbook
tf	140.50 ± 0.50	K	NIST Webbook

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	213.75	J/molxK	393.05	Joback Method
cpg	230.45	J/molxK	425.08	Joback Method
cpg	246.41	J/molxK	457.10	Joback Method
cpg	261.64	J/molxK	489.13	Joback Method
cpg	276.17	J/molxK	521.15	Joback Method
cpg	290.01	J/molxK	553.18	Joback Method
cpg	303.19	J/molxK	585.20	Joback Method
dvisc	0.0022106	Paxs	186.58	Joback Method
dvisc	0.0012051	Paxs	220.99	Joback Method
dvisc	0.0007737	Paxs	255.40	Joback Method
dvisc	0.0005518	Paxs	289.81	Joback Method
dvisc	0.0004228	Paxs	324.23	Joback Method
dvisc	0.0003410	Paxs	358.64	Joback Method
dvisc	0.0002855	Paxs	393.05	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.35282e+01
Coeff. B	-3.06019e+03
Coeff. C	-5.07380e+01
Temperature range (K), min.	281.86
Temperature range (K), max.	423.17

## Sources

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C3726474&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C3726474&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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