

# 4,4'-Diisopropylbiphenyl

<b>Other names:</b>	1,1'-Biphenyl, 4,4'-bis-(1-methylethyl) 1,1'-Biphenyl, 4,4'-diisopropyl p,p'-Diisopropylbiphenyl Biphenyl, 4,4'-diisopropyl- 3,4'-Diisopropylbiphenyl
<b>Inchi:</b>	InChI=1S/C18H22/c1-13(2)15-5-9-17(10-6-15)18-11-7-16(8-12-18)14(3)4/h5-14H,1-4H3
<b>InchiKey:</b>	NUEUMFZLNOCRCQ-UHFFFAOYSA-N
<b>Formula:</b>	C18H22
<b>SMILES:</b>	CC(C)c1ccc(-c2ccc(C(C)C)cc2)cc1
<b>Mol. weight [g/mol]:</b>	238.37
<b>CAS:</b>	18970-30-4

## Physical Properties

Property code	Value	Unit	Source
gf	301.36	kJ/mol	Joback Method
hf	24.71	kJ/mol	Joback Method
hfus	22.63	kJ/mol	Joback Method
hvap	60.76	kJ/mol	Joback Method
log10ws	-6.44		Crippen Method
logp	5.600		Crippen Method
mcvol	216.960	ml/mol	McGowan Method
pc	1893.65	kPa	Joback Method
rinpol	1963.00		NIST Webbook
rinpol	1963.00		NIST Webbook
tb	608.00 ± 10.00	K	NIST Webbook
tc	905.93	K	Joback Method
tf	322.00 ± 3.00	K	NIST Webbook
tf	338.00 ± 20.00	K	NIST Webbook
tf	322.00 ± 2.00	K	NIST Webbook
tf	339.00 ± 4.00	K	NIST Webbook
vc	0.816	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	578.18	J/mol×K	673.68	Joback Method
cpg	677.08	J/mol×K	905.93	Joback Method
cpg	663.54	J/mol×K	867.22	Joback Method
cpg	648.92	J/mol×K	828.51	Joback Method
cpg	633.14	J/mol×K	789.80	Joback Method
cpg	616.14	J/mol×K	751.10	Joback Method
cpg	597.85	J/mol×K	712.39	Joback Method
cpl	520.50	J/mol×K	422.00	NIST Webbook
dvisc	0.0001049	Paxs	673.68	Joback Method
dvisc	0.0001378	Paxs	618.15	Joback Method
dvisc	0.0001911	Paxs	562.62	Joback Method
dvisc	0.0002845	Paxs	507.09	Joback Method
dvisc	0.0004672	Paxs	451.56	Joback Method
dvisc	0.0008817	Paxs	396.03	Joback Method
dvisc	0.0020468	Paxs	340.50	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C18970304&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C18970304&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>cpl:</b>	Liquid phase heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure

<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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