

# 2-tert-Butyl-4-methyl-6-( $\alpha$ -methylbenzyl)phenol

<b>Other names:</b>	2-tert-butyl-4-methyl-6- $\alpha$ -methylbenzylphenol
<b>Inchi:</b>	InChI=1S/C19H24O/c1-13-11-16(14(2)15-9-7-6-8-10-15)18(20)17(12-13)19(3,4)5/h6-12,
<b>InchiKey:</b>	KYWXWCOPNNMDMN-UHFFFAOYSA-N
<b>Formula:</b>	C19H24O
<b>SMILES:</b>	<chem>Cc1cc(C(C)c2ccccc2)c(O)c(C(C)(C)C)c1</chem>
<b>Mol. weight [g/mol]:</b>	268.39
<b>CAS:</b>	1706-65-6

## Physical Properties

Property code	Value	Unit	Source
gf	160.44	kJ/mol	Joback Method
hf	-176.71	kJ/mol	Joback Method
hfus	27.12	kJ/mol	Joback Method
hvap	75.09	kJ/mol	Joback Method
log10ws	-5.29		Crippen Method
logp	5.150		Crippen Method
mcvol	236.920	ml/mol	McGowan Method
pc	2003.70	kPa	Joback Method
tb	774.39	K	Joback Method
tc	1017.45	K	Joback Method
tf	338.00 $\pm$ 0.20	K	NIST Webbook
vc	0.833	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	786.77	J/molxK	1017.45	Joback Method
cpg	772.92	J/molxK	976.94	Joback Method
cpg	758.61	J/molxK	936.43	Joback Method
cpg	743.69	J/molxK	895.92	Joback Method
cpg	727.99	J/molxK	855.41	Joback Method
cpg	711.33	J/molxK	814.90	Joback Method
cpg	693.55	J/molxK	774.39	Joback Method
dvisc	0.0002527	Paxs	480.91	Joback Method

dvisc	0.0000055	Paxs	774.39	Joback Method
dvisc	0.0000084	Paxs	725.48	Joback Method
dvisc	0.0000136	Paxs	676.56	Joback Method
dvisc	0.0000239	Paxs	627.65	Joback Method
dvisc	0.0000459	Paxs	578.74	Joback Method
dvisc	0.0000995	Paxs	529.82	Joback Method
hfust	31.38	kJ/mol	337.70	NIST Webbook

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1706656&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1706656&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/58-417-5/2-tert-Butyl-4-methyl-6-a-methylbenzyl-phenol.pdf>

Generated by Cheméo on 2024-04-26 17:38:04.224848792 +0000 UTC m=+16442333.145426114.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.