

# Octanoic acid, 3-hydroxy-, methyl ester

<b>Other names:</b>	methyl 3-hydroxyoctanoate
<b>Inchi:</b>	InChI=1S/C9H18O3/c1-3-4-5-6-8(10)7-9(11)12-2/h8,10H,3-7H2,1-2H3
<b>InchiKey:</b>	FHWBTAQR RDZDIY-UHFFFAOYSA-N
<b>Formula:</b>	C9H18O3
<b>SMILES:</b>	CCCCC(O)CC(=O)OC
<b>Mol. weight [g/mol]:</b>	174.24
<b>CAS:</b>	7367-87-5

## Physical Properties

Property code	Value	Unit	Source
gf	-348.28	kJ/mol	Joback Method
hf	-631.40	kJ/mol	Joback Method
hfus	22.42	kJ/mol	Joback Method
hvap	61.08	kJ/mol	Joback Method
log10ws	-1.83		Crippen Method
logp	1.491		Crippen Method
mcvol	150.980	ml/mol	McGowan Method
pc	2668.02	kPa	Joback Method
rinpol	1243.00		NIST Webbook
rinpol	1243.00		NIST Webbook
ripol	1850.00		NIST Webbook
ripol	1835.00		NIST Webbook
ripol	1835.00		NIST Webbook
ripol	1807.00		NIST Webbook
ripol	1850.00		NIST Webbook
tb	573.35	K	Joback Method
tc	743.84	K	Joback Method
tf	309.17	K	Joback Method
vc	0.577	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	380.44	J/mol×K	573.35	Joback Method

cpg	392.26	J/molxK	601.77	Joback Method
cpg	403.60	J/molxK	630.18	Joback Method
cpg	414.48	J/molxK	658.60	Joback Method
cpg	424.88	J/molxK	687.01	Joback Method
cpg	434.83	J/molxK	715.43	Joback Method
cpg	444.32	J/molxK	743.84	Joback Method
dvisc	0.0115627	Paxs	309.17	Joback Method
dvisc	0.0030766	Paxs	353.20	Joback Method
dvisc	0.0010979	Paxs	397.23	Joback Method
dvisc	0.0004812	Paxs	441.26	Joback Method
dvisc	0.0002450	Paxs	485.29	Joback Method
dvisc	0.0001395	Paxs	529.32	Joback Method
dvisc	0.0000867	Paxs	573.35	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C7367875&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C7367875&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

**tf:** Normal melting (fusion) point

**vc:** Critical Volume

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