

# Methyl ethyl cyclopentene

<b>Other names:</b>	1-Ethyl-2-methylcyclopentene Cyclopentene, 1-ethyl-2-methyl-
<b>Inchi:</b>	InChI=1S/C8H14/c1-3-8-6-4-5-7(8)2/h3-6H2,1-2H3
<b>InchiKey:</b>	MMYZGQDXYZNAQW-UHFFFAOYSA-N
<b>Formula:</b>	C8H14
<b>SMILES:</b>	CCC1=C(C)CCC1
<b>Mol. weight [g/mol]:</b>	110.20
<b>CAS:</b>	19780-56-4

## Physical Properties

Property code	Value	Unit	Source
gf	71.44	kJ/mol	Joback Method
hf	-92.79	kJ/mol	Joback Method
hfus	9.78	kJ/mol	Joback Method
hvap	35.58	kJ/mol	Joback Method
log10ws	-2.92		Crippen Method
logp	2.897		Crippen Method
mvol	108.420	ml/mol	McGowan Method
pc	3202.78	kPa	Joback Method
tb	411.51	K	Joback Method
tc	609.80	K	Joback Method
tf	220.86	K	Joback Method
vc	0.411	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	204.62	J/molxK	411.51	Joback Method
cpg	218.58	J/molxK	444.56	Joback Method
cpg	231.87	J/molxK	477.61	Joback Method
cpg	244.51	J/molxK	510.65	Joback Method
cpg	256.53	J/molxK	543.70	Joback Method
cpg	267.96	J/molxK	576.75	Joback Method
cpg	278.81	J/molxK	609.80	Joback Method

dvisc	0.0023108	Paxs	220.86	Joback Method
dvisc	0.0012898	Paxs	252.64	Joback Method
dvisc	0.0008200	Paxs	284.41	Joback Method
dvisc	0.0005711	Paxs	316.19	Joback Method
dvisc	0.0004249	Paxs	347.96	Joback Method
dvisc	0.0003321	Paxs	379.74	Joback Method
dvisc	0.0002697	Paxs	411.51	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C19780564&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C19780564&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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