

# 4-Chloro-4'-nitrodiphenyl sulfide

<b>Other names:</b>	4-Nitro-4'-chloro diphenyl sulfide Benzene, 1-chloro-4-[(4-nitrophenyl)thio]-
<b>Inchi:</b>	InChI=1S/C12H8ClNO2S/c13-9-1-5-11(6-2-9)17-12-7-3-10(4-8-12)14(15)16/h1-8H
<b>InchiKey:</b>	CPUBRDUBQIVBGM-UHFFFAOYSA-N
<b>Formula:</b>	C12H8ClNO2S
<b>SMILES:</b>	O=[N+][O-]c1ccc(Sc2ccc(Cl)cc2)cc1
<b>Mol. weight [g/mol]:</b>	265.71
<b>CAS:</b>	21969-11-9

## Physical Properties

Property code	Value	Unit	Source
gf	312.46	kJ/mol	Joback Method
hf	174.48	kJ/mol	Joback Method
hfus	33.83	kJ/mol	Joback Method
hvap	75.97	kJ/mol	Joback Method
log10ws	-5.22		Crippen Method
logp	4.399		Crippen Method
mvol	178.430	ml/mol	McGowan Method
pc	3280.28	kPa	Joback Method
tb	795.33	K	Joback Method
tc	1088.59	K	Joback Method
tf	510.81	K	Joback Method
vc	0.676	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	438.08	J/molxK	795.33	Joback Method
cpg	449.23	J/molxK	844.21	Joback Method
cpg	459.05	J/molxK	893.08	Joback Method
cpg	467.64	J/molxK	941.96	Joback Method
cpg	475.10	J/molxK	990.84	Joback Method
cpg	481.52	J/molxK	1039.71	Joback Method
cpg	486.99	J/molxK	1088.59	Joback Method

# Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C21969119&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C21969119&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvp:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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