

# p-menth-2-en-7-ol

<b>Other names:</b>	Menth-2-en-7-ol
<b>Inchi:</b>	InChI=1S/C10H18O/c1-8(2)10-5-3-9(7-11)4-6-10/h3,5,8-11H,4,6-7H2,1-2H3
<b>InchiKey:</b>	JWRJGMUBDGIGRT-UHFFFAOYSA-N
<b>Formula:</b>	C10H18O
<b>SMILES:</b>	CC(C)C1C=CC(CO)CC1
<b>Mol. weight [g/mol]:</b>	154.25

## Physical Properties

Property code	Value	Unit	Source
gf	-59.24	kJ/mol	Joback Method
hf	-315.48	kJ/mol	Joback Method
hfus	16.35	kJ/mol	Joback Method
hvap	54.56	kJ/mol	Joback Method
log10ws	-2.30		Crippen Method
logp	2.217		Crippen Method
mcvol	142.470	ml/mol	McGowan Method
pc	2859.68	kPa	Joback Method
rinpol	1199.00		NIST Webbook
rinpol	1199.00		NIST Webbook
ripol	1778.00		NIST Webbook
ripol	1760.00		NIST Webbook
ripol	1743.00		NIST Webbook
tb	533.98	K	Joback Method
tc	725.42	K	Joback Method
tf	252.18	K	Joback Method
vc	0.526	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	351.36	J/molxK	533.98	Joback Method
cpg	367.42	J/molxK	565.89	Joback Method
cpg	382.68	J/molxK	597.79	Joback Method
cpg	397.16	J/molxK	629.70	Joback Method

cpg	410.88	J/molxK	661.61	Joback Method
cpg	423.86	J/molxK	693.52	Joback Method
cpg	436.13	J/molxK	725.42	Joback Method
dvisc	0.0448779	Paxs	252.18	Joback Method
dvisc	0.0078994	Paxs	299.15	Joback Method
dvisc	0.0022279	Paxs	346.11	Joback Method
dvisc	0.0008503	Paxs	393.08	Joback Method
dvisc	0.0003986	Paxs	440.05	Joback Method
dvisc	0.0002163	Paxs	487.01	Joback Method
dvisc	0.0001307	Paxs	533.98	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R219235&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R219235&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/58-499-5/p-menth-2-en-7-ol.pdf>

Generated by Cheméo on 2024-04-29 13:44:51.917326303 +0000 UTC m=+16687540.837903614.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.