

Isophthalic acid, hexyl hex-3-yl ester

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| Inchi: | InChI=1S/C20H30O4/c1-4-7-8-9-14-23-19(21)16-12-10-13-17(15-16)20(22)24-18(6-3)11 |
| InchiKey: | CJOYNWJJQNCTOU-UHFFFAOYSA-N |
| Formula: | C20H30O4 |
| SMILES: | CCCCCOC(=O)c1cccc(C(=O)OC(CC)CCC)c1 |
| Mol. weight [g/mol]: | 334.45 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|----------------------|----------------|
| gf | -249.98 | kJ/mol | Joback Method |
| hf | -725.95 | kJ/mol | Joback Method |
| hfus | 43.26 | kJ/mol | Joback Method |
| hvap | 80.98 | kJ/mol | Joback Method |
| log10ws | -6.25 | | Crippen Method |
| logp | 5.159 | | Crippen Method |
| mvol | 283.780 | ml/mol | McGowan Method |
| pc | 1349.66 | kPa | Joback Method |
| rinpol | 2441.00 | | NIST Webbook |
| tb | 840.80 | K | Joback Method |
| tc | 1042.55 | K | Joback Method |
| tf | 483.42 | K | Joback Method |
| vc | 1.089 | m ³ /kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|-----------|---------|-----------------|---------------|
| cpg | 878.24 | J/molxK | 840.80 | Joback Method |
| cpg | 948.18 | J/molxK | 1008.93 | Joback Method |
| cpg | 936.41 | J/molxK | 975.30 | Joback Method |
| cpg | 923.56 | J/molxK | 941.68 | Joback Method |
| cpg | 909.59 | J/molxK | 908.05 | Joback Method |
| cpg | 894.49 | J/molxK | 874.43 | Joback Method |
| cpg | 958.87 | J/molxK | 1042.55 | Joback Method |
| dvisc | 0.0000502 | Paxs | 840.80 | Joback Method |
| dvisc | 0.0000656 | Paxs | 781.24 | Joback Method |

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|-------|-----------|------|--------|---------------|
| dvisc | 0.0000897 | Paxs | 721.67 | Joback Method |
| dvisc | 0.0001295 | Paxs | 662.11 | Joback Method |
| dvisc | 0.0002012 | Paxs | 602.55 | Joback Method |
| dvisc | 0.0003442 | Paxs | 542.98 | Joback Method |
| dvisc | 0.0006723 | Paxs | 483.42 | Joback Method |

Sources

| | |
|------------------------|---|
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=U356489&Units=SI |
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci990307l |
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |

Legend

| | |
|-----------------|---|
| cpg: | Ideal gas heat capacity |
| dvisc: | Dynamic viscosity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| rinpol: | Non-polar retention indices |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

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