

# 3-ethyl-indole

<b>Other names:</b>	3-ethyl-1H-indole
<b>Inchi:</b>	InChI=1S/C10H11N/c1-2-8-7-11-10-6-4-3-5-9(8)10/h3-7,11H,2H2,1H3
<b>InchiKey:</b>	GOVXKUCVZUROAN-UHFFFAOYSA-N
<b>Formula:</b>	C10H11N
<b>SMILES:</b>	CCc1c[nH]c2ccccc12
<b>Mol. weight [g/mol]:</b>	145.20
<b>CAS:</b>	1484-19-1

## Physical Properties

Property code	Value	Unit	Source
log10ws	-3.30		Crippen Method
logp	2.248		Crippen Method
mccvol	122.820	ml/mol	McGowan Method
rinpol	1493.00		NIST Webbook
rinpol	1479.00		NIST Webbook
rinpol	1479.00		NIST Webbook
ripol	2509.00		NIST Webbook
ripol	2540.00		NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.43476e+01
Coeff. B	-4.50302e+03
Coeff. C	-9.46480e+01
Temperature range (K), min.	414.92
Temperature range (K), max.	592.99

# Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1484191&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1484191&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

# Legend

<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices

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