

# 1-Phenanthrenecarboxylic acid, 1,2,3,4,4a,4b,5,6,7,9,10,10a-dodecahydro-1,4a-dimethano- [1R-(1«alpha»,4a«beta»,4b«alpha»,10a«alpha»)]-

Other names:

Podocarp-8(14)-en-15-oiic acid, 13-isopropylidene-  
acid  
Neoabietic acid

**Inchi:** InChI=1S/C20H30O2/c1-13(2)14-6-8-16-15(12-14)7-9-17-19(16,3)10-5-11-20(17,4)18(2)1-20  
**InchiKey:** KGMSWPSAVZAMKR-UHFFFAOYSA-N  
**Formula:** C20H30O2  
**SMILES:** CC(C)=C1C=C2CCC3C(C)(C(=O)O)CCCC3(C)C2CC1  
**Mol. weight [g/mol]:** 302.45  
**CAS:** 471-77-2

## Physical Properties

Property code	Value	Unit	Source
gf	12.08	kJ/mol	Joback Method
hf	-410.65	kJ/mol	Joback Method
hfus	25.47	kJ/mol	Joback Method
hvap	83.35	kJ/mol	Joback Method
log10ws	-5.72		Crippen Method
logp	5.350		Crippen Method
mcvol	258.920	ml/mol	McGowan Method
pc	1801.56	kPa	Joback Method
rinpol	2335.00		NIST Webbook
rinpol	2335.00		NIST Webbook
tb	851.09	K	Joback Method
tc	1076.87	K	Joback Method
tf	515.37	K	Joback Method
vc	0.977	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	868.76	J/molxK	851.09	Joback Method
cpg	891.18	J/molxK	888.72	Joback Method
cpg	913.79	J/molxK	926.35	Joback Method
cpg	936.86	J/molxK	963.98	Joback Method

cpg	960.71	J/mol×K	1001.61	Joback Method
cpg	985.62	J/mol×K	1039.24	Joback Method
cpg	1011.90	J/mol×K	1076.87	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C471772&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C471772&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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