

# Glutaric acid, 1-phenylpropyl propyl ester

<b>Inchi:</b>	InChI=1S/C17H24O4/c1-3-13-20-16(18)11-8-12-17(19)21-15(4-2)14-9-6-5-7-10-14/h5-7,
<b>InchiKey:</b>	ZDJAZZJWVAFNTF-UHFFFAOYSA-N
<b>Formula:</b>	C17H24O4
<b>SMILES:</b>	CCCOC(=O)CCCC(=O)OC(CC)c1ccccc1
<b>Mol. weight [g/mol]:</b>	292.37

## Physical Properties

Property code	Value	Unit	Source
gf	-265.61	kJ/mol	Joback Method
hf	-652.56	kJ/mol	Joback Method
hfus	35.88	kJ/mol	Joback Method
hvap	73.64	kJ/mol	Joback Method
log10ws	-4.23		Crippen Method
logp	3.804		Crippen Method
mcvol	241.510	ml/mol	McGowan Method
pc	1708.95	kPa	Joback Method
rinpola	2075.00		NIST Webbook
tb	767.18	K	Joback Method
tc	969.48	K	Joback Method
tf	437.09	K	Joback Method
vc	0.921	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	705.62	J/molxK	767.18	Joback Method
cpg	721.35	J/molxK	800.90	Joback Method
cpg	736.03	J/molxK	834.61	Joback Method
cpg	749.67	J/molxK	868.33	Joback Method
cpg	762.30	J/molxK	902.04	Joback Method
cpg	773.93	J/molxK	935.76	Joback Method
cpg	784.60	J/molxK	969.48	Joback Method
dvisc	0.0010935	Paxs	437.09	Joback Method
dvisc	0.0005455	Paxs	492.11	Joback Method

dvisc	0.0003130	Paxs	547.12	Joback Method
dvisc	0.0001988	Paxs	602.13	Joback Method
dvisc	0.0001362	Paxs	657.15	Joback Method
dvisc	0.0000989	Paxs	712.16	Joback Method
dvisc	0.0000752	Paxs	767.18	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U358948&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U358948&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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