

# Ethane, 1,2-dicyclopropyl-

<b>Other names:</b>	Cyclopropane, 1,1'-(1,2-ethanediyl)bis-(2-Cyclopropylethyl)cyclopropane (3,4-Methylene)butyl-cyclopropane
<b>Inchi:</b>	InChI=1S/C8H14/c1-2-7(1)5-6-8-3-4-8/h7-8H,1-6H2
<b>InchiKey:</b>	KMEBZBOYYXYOIU-UHFFFAOYSA-N
<b>Formula:</b>	C8H14
<b>SMILES:</b>	C1CC1CCC1CC1
<b>Mol. weight [g/mol]:</b>	110.20
<b>CAS:</b>	37520-11-9

## Physical Properties

Property code	Value	Unit	Source
gf	137.98	kJ/mol	Joback Method
hf	-62.85	kJ/mol	Joback Method
hfus	12.75	kJ/mol	Joback Method
hvap	33.23	kJ/mol	Joback Method
log10ws	-2.48		Crippen Method
logp	2.587		Crippen Method
mcvol	101.860	ml/mol	McGowan Method
pc	3356.75	kPa	Joback Method
rinsol	826.80		NIST Webbook
rinsol	826.80		NIST Webbook
tb	395.92	K	Joback Method
tc	589.01	K	Joback Method
tf	215.80	K	Joback Method
vc	0.398	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	201.04	J/molxK	395.92	Joback Method
cpg	273.17	J/molxK	556.83	Joback Method
cpg	260.54	J/molxK	524.65	Joback Method
cpg	247.07	J/molxK	492.47	Joback Method

cpg	232.71	J/molxK	460.28	Joback Method
cpg	217.38	J/molxK	428.10	Joback Method
cpg	285.01	J/molxK	589.01	Joback Method
dvisc	0.0005493	Paxs	395.92	Joback Method
dvisc	0.0005530	Paxs	365.90	Joback Method
dvisc	0.0005574	Paxs	335.88	Joback Method
dvisc	0.0005626	Paxs	305.86	Joback Method
dvisc	0.0005690	Paxs	275.84	Joback Method
dvisc	0.0005772	Paxs	245.82	Joback Method
dvisc	0.0005877	Paxs	215.80	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C37520119&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C37520119&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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