

# Tritetracontane

**Inchi:** InChI=1S/C43H88/c1-3-5-7-9-11-13-15-17-19-21-23-25-27-29-31-33-35-37-39-41-43-42-43-45-47-49-51-53-55-57-59-61-63-65-67-69-71-73-75-77-79-81-83-85-87-89-91-93-95-97-99-100  
**InchiKey:** PKBVGJFKBPLWOD-UHFFFAOYSA-N  
**Formula:** C43H88  
**SMILES:** CCC  
**Mol. weight [g/mol]:** 605.16  
**CAS:** 7098-21-7

## Physical Properties

Property code	Value	Unit	Source
gf	311.18	kJ/mol	Joback Method
hf	-930.85	kJ/mol	Joback Method
hfus	107.13	kJ/mol	Joback Method
hvap	111.31	kJ/mol	Joback Method
log10ws	-17.82		Crippen Method
logp	17.020		Crippen Method
mcvol	616.730	ml/mol	McGowan Method
pc	349.90	kPa	Joback Method
tb	1183.24	K	Joback Method
tc	1657.15	K	Joback Method
tf	358.45 ± 0.25	K	NIST Webbook
tf	357.00 ± 2.00	K	NIST Webbook
tf	358.20 ± 2.00	K	NIST Webbook
tf	358.20 ± 1.00	K	NIST Webbook
tf	357.70 ± 2.00	K	NIST Webbook
tf	357.00 ± 4.00	K	NIST Webbook
tf	356.50 ± 4.00	K	NIST Webbook
vc	2.443	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	2389.32	J/mol×K	1183.24	Joback Method
cpg	2617.18	J/mol×K	1578.16	Joback Method
cpg	2576.17	J/mol×K	1499.18	Joback Method

cpg	2534.36	J/mol×K	1420.19	Joback Method
cpg	2490.27	J/mol×K	1341.21	Joback Method
cpg	2442.42	J/mol×K	1262.22	Joback Method
cpg	2658.86	J/mol×K	1657.15	Joback Method
dvisc	0.0000030	Paxs	1183.24	Joback Method
dvisc	0.0000043	Paxs	1081.76	Joback Method
dvisc	0.0000065	Paxs	980.28	Joback Method
dvisc	0.0000108	Paxs	878.80	Joback Method
dvisc	0.0000207	Paxs	777.33	Joback Method
dvisc	0.0000480	Paxs	675.85	Joback Method
dvisc	0.0001502	Paxs	574.37	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.40245e+01
Coeff. B	-5.46721e+03
Coeff. C	-2.43550e+02
Temperature range (K), min.	641.54
Temperature range (K), max.	871.03

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C7098217&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C7098217&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity

<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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