

# 1,2-Difluorotetranitroethane

**Inchi:** InChI=1S/C2F2N4O8/c3-1(5(9)10,6(11)12)2(4,7(13)14)8(15)16  
**InchiKey:** BDYYJOSEHYEBKF-UHFFFAOYSA-N  
**Formula:** C2F2N4O8  
**SMILES:** O=[N+]([O-])C(F)([N+](=O)[O-])C(F)([N+](=O)[O-])[N+](=O)[O-]  
**Mol. weight [g/mol]:** 246.04  
**CAS:** 20165-39-3

## Physical Properties

Property code	Value	Unit	Source
chl	-730.00 ± 3.00	kJ/mol	NIST Webbook
gf	-275.78	kJ/mol	Joback Method
hf	-352.00 ± 4.20	kJ/mol	NIST Webbook
hfl	-415.00 ± 4.20	kJ/mol	NIST Webbook
hfus	37.71	kJ/mol	Joback Method
hvap	62.80 ± 0.80	kJ/mol	NIST Webbook
log10ws	-2.91		Crippen Method
logp	-0.660		Crippen Method
mcvol	112.260	ml/mol	McGowan Method
pc	5001.51	kPa	Joback Method
tb	844.60	K	Joback Method
tc	1137.10	K	Joback Method
tf	692.76	K	Joback Method
vc	0.489	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	313.92	J/molxK	844.60	Joback Method
cpg	317.76	J/molxK	893.35	Joback Method
cpg	321.06	J/molxK	942.10	Joback Method
cpg	323.99	J/molxK	990.85	Joback Method
cpg	326.72	J/molxK	1039.60	Joback Method
cpg	329.43	J/molxK	1088.35	Joback Method
cpg	332.29	J/molxK	1137.10	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C20165393&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C20165393&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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