

# 2-Propenoyl chloride

<b>Other names:</b>	Acryloyl chloride Acrylic acid chloride Acrylyl chloride Propenoyl chloride Chlorid kyseliny akrylove
<b>Inchi:</b>	InChI=1S/C3H3ClO/c1-2-3(4)5/h2H,1H2
<b>InchiKey:</b>	HFBMWMNUJJDEQZ-UHFFFAOYSA-N
<b>Formula:</b>	C3H3ClO
<b>SMILES:</b>	C=CC(=O)Cl
<b>Mol. weight [g/mol]:</b>	90.51
<b>CAS:</b>	814-68-6

## Physical Properties

Property code	Value	Unit	Source
gf	-78.63	kJ/mol	Joback Method
hf	-108.14	kJ/mol	Joback Method
hfus	8.04	kJ/mol	Joback Method
hvap	32.73	kJ/mol	Joback Method
log10ws	-0.86		Crippen Method
logp	0.938		Crippen Method
mcvol	62.640	ml/mol	McGowan Method
pc	4835.95	kPa	Joback Method
tb	348.70	K	NIST Webbook
tb	347.20	K	NIST Webbook
tb	348.00	K	NIST Webbook
tc	548.68	K	Joback Method
tf	201.66	K	Joback Method
vc	0.239	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	86.67	J/mol×K	356.02	Joback Method
cpg	91.26	J/mol×K	388.13	Joback Method

cpg	95.62	J/mol×K	420.24	Joback Method
cpg	99.74	J/mol×K	452.35	Joback Method
cpg	103.64	J/mol×K	484.46	Joback Method
cpg	107.33	J/mol×K	516.57	Joback Method
cpg	110.82	J/mol×K	548.68	Joback Method
dvisc	0.0025952	Paxs	201.66	Joback Method
dvisc	0.0015270	Paxs	227.39	Joback Method
dvisc	0.0010008	Paxs	253.11	Joback Method
dvisc	0.0007091	Paxs	278.84	Joback Method
dvisc	0.0005325	Paxs	304.57	Joback Method
dvisc	0.0004182	Paxs	330.29	Joback Method
dvisc	0.0003400	Paxs	356.02	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C814686&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C814686&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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