

# Peroxide, dibutyl

<b>Other names:</b>	Butyl peroxide Dibutyl peroxide (n-C <sub>4</sub> H <sub>9</sub> O) <sub>2</sub>
<b>Inchi:</b>	InChI=1S/C <sub>8</sub> H <sub>18</sub> O <sub>2</sub> /c1-3-5-7-9-10-8-6-4-2/h3-8H2,1-2H3
<b>InchiKey:</b>	PAOHAQSLJSMLAT-UHFFFAOYSA-N
<b>Formula:</b>	C <sub>8</sub> H <sub>18</sub> O <sub>2</sub>
<b>SMILES:</b>	CCCCOCCCC
<b>Mol. weight [g/mol]:</b>	146.23
<b>CAS:</b>	3849-34-1

## Physical Properties

Property code	Value	Unit	Source
chl	-5326.20 ± 5.00	kJ/mol	NIST Webbook
gf	-293.89	kJ/mol	Joback Method
hf	-525.73	kJ/mol	Joback Method
hfus	20.85	kJ/mol	Joback Method
hvap	40.00	kJ/mol	NIST Webbook
log10ws	-2.34		Crippen Method
logp	2.535		Crippen Method
mcvol	135.320	ml/mol	McGowan Method
pc	2391.19	kPa	Joback Method
tb	440.66	K	Joback Method
tc	612.33	K	Joback Method
tf	222.25	K	Joback Method
vc	0.509	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	274.26	J/mol×K	440.66	Joback Method
cpg	286.77	J/mol×K	469.27	Joback Method
cpg	298.83	J/mol×K	497.88	Joback Method
cpg	310.44	J/mol×K	526.49	Joback Method
cpg	321.60	J/mol×K	555.10	Joback Method

cpg	332.31	J/mol×K	583.71	Joback Method
cpg	342.58	J/mol×K	612.33	Joback Method
dvisc	0.0062439	Paxs	222.25	Joback Method
dvisc	0.0024874	Paxs	258.65	Joback Method
dvisc	0.0012435	Paxs	295.05	Joback Method
dvisc	0.0007239	Paxs	331.45	Joback Method
dvisc	0.0004691	Paxs	367.86	Joback Method
dvisc	0.0003287	Paxs	404.26	Joback Method
dvisc	0.0002442	Paxs	440.66	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C3849341&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C3849341&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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