

1-Tritriacontene

Inchi:	InChI=1S/C33H66/c1-3-5-7-9-11-13-15-17-19-21-23-25-27-29-31-33-32-30-28-26-24-22
InchiKey:	DRJAAODMPBOZLH-UHFFFAOYSA-N
Formula:	C33H66
SMILES:	C=CCCCCCCCCCCCCCCCCCCCCCCCCCCCCCCCC
Mol. weight [g/mol]:	462.88
CAS:	61868-11-9

Physical Properties

Property code	Value	Unit	Source
gf	314.82	kJ/mol	Joback Method
hf	-599.02	kJ/mol	Joback Method
hfus	79.95	kJ/mol	Joback Method
hvap	88.38	kJ/mol	Joback Method
log10ws	-13.49		Crippen Method
logp	12.895		Crippen Method
mcvol	471.530	ml/mol	McGowan Method
pc	529.93	kPa	Joback Method
rinpol	3288.00		NIST Webbook
rinpol	3288.00		NIST Webbook
tb	951.12	K	Joback Method
tc	1186.34	K	Joback Method
tf	459.91	K	Joback Method
vc	1.865	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1656.31	J/molxK	951.12	Joback Method
cpg	1686.23	J/molxK	990.32	Joback Method
cpg	1714.27	J/molxK	1029.53	Joback Method
cpg	1740.59	J/molxK	1068.73	Joback Method
cpg	1765.32	J/molxK	1107.93	Joback Method
cpg	1788.61	J/molxK	1147.14	Joback Method
cpg	1810.60	J/molxK	1186.34	Joback Method

dvisc	0.0007138	Paxs	459.91	Joback Method
dvisc	0.0002372	Paxs	541.78	Joback Method
dvisc	0.0001053	Paxs	623.65	Joback Method
dvisc	0.0000564	Paxs	705.51	Joback Method
dvisc	0.0000344	Paxs	787.38	Joback Method
dvisc	0.0000231	Paxs	869.25	Joback Method
dvisc	0.0000165	Paxs	951.12	Joback Method

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.51171e+01
Coeff. B	-6.31499e+03
Coeff. C	-1.48580e+02
Temperature range (K), min.	574.42
Temperature range (K), max.	792.60

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=R250776&Units=SI
The Yaws Handbook of Vapor Pressure:	https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions

hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
pvap:	Vapor pressure
rinpola:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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