

3,4-DICHLOROBENZAMIDE

Inchi:	InChI=1S/C7H5Cl2NO/c8-5-2-1-4(7(10)11)3-6(5)9/h1-3H,(H2,10,11)
InchiKey:	XURBWYCGQQXTHJ-UHFFFAOYSA-N
Formula:	C7H5Cl2NO
SMILES:	NC(=O)c1ccc(Cl)c(Cl)c1
Mol. weight [g/mol]:	190.03

Physical Properties

Property code	Value	Unit	Source
hf	-215.71	kJ/mol	Cheméo Relay (1.0)
hfus	22.34	kJ/mol	Joback Method
hvap	78.17	kJ/mol	Cheméo Relay (1.0)
ie	9.43	eV	Cheméo Relay (1.0)
log10ws	-3.24		Cheméo Relay (1.0)
logp	2.092		Crippen Method
mcvol	121.760	ml/mol	McGowan Method
tb	559.11	K	Cheméo Relay (1.0)
tf	382.14	K	Cheméo Relay (1.0)

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	240.93	J/mol×K	597.46	Joback Method
cpg	249.24	J/mol×K	638.77	Joback Method
cpg	256.89	J/mol×K	680.08	Joback Method
cpg	263.93	J/mol×K	721.39	Joback Method
cpg	270.37	J/mol×K	762.70	Joback Method
cpg	276.25	J/mol×K	804.01	Joback Method
cpg	281.60	J/mol×K	845.33	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Cheméo Relay (1.0):	
Cheméo Relay (1.0, beta):	
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C2670384&Units=SI
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
ie:	Ionization energy
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
tb:	Normal Boiling Point Temperature
tf:	Normal melting (fusion) point

Latest version available from:

<https://www.chemeo.com/cid/58-872-0/3-4-DICHLOROBENZAMIDE.pdf>

Generated by Cheméo on 2026-07-21 20:09:25.398711116 +0000 UTC m=+175661.240596482.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.