

# (3-Chloro-4-cyclohexylmethoxy-phenyl)-acetic acid, methyl ester

Inchi:	InChI=1S/C16H21ClO3/c1-19-16(18)10-13-7-8-15(14(17)9-13)20-11-12-5-3-2-4-6-12/h7-
InchiKey:	KXICENYJWTTXKL-UHFFFAOYSA-N
Formula:	C16H21ClO3
SMILES:	COC(=O)Cc1ccc(OCC2CCCCC2)c(Cl)c1
Mol. weight [g/mol]:	296.79

## Physical Properties

Property code	Value	Unit	Source
gf	-149.41	kJ/mol	Joback Method
hf	-498.42	kJ/mol	Joback Method
hfus	30.47	kJ/mol	Joback Method
hvap	71.19	kJ/mol	Joback Method
log10ws	-4.52		Crippen Method
logp	4.015		Crippen Method
mcvol	227.230	ml/mol	McGowan Method
pc	1964.82	kPa	Joback Method
rinsol	2224.70		NIST Webbook
tb	757.81	K	Joback Method
tc	985.53	K	Joback Method
tf	453.23	K	Joback Method
vc	0.848	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	650.92	J/molxK	757.81	Joback Method
cpg	724.90	J/molxK	947.58	Joback Method
cpg	712.76	J/molxK	909.63	Joback Method
cpg	699.31	J/molxK	871.67	Joback Method
cpg	684.53	J/molxK	833.72	Joback Method
cpg	668.41	J/molxK	795.76	Joback Method
cpg	735.74	J/molxK	985.53	Joback Method
dvisc	0.0000890	Paxs	757.81	Joback Method
dvisc	0.0001131	Paxs	707.05	Joback Method

dvisc	0.0001492	Paxs	656.28	Joback Method
dvisc	0.0002063	Paxs	605.52	Joback Method
dvisc	0.0003025	Paxs	554.76	Joback Method
dvisc	0.0004792	Paxs	503.99	Joback Method
dvisc	0.0008415	Paxs	453.23	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R157974&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R157974&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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