

# Androstane-3,17-diol, (3«alpha»,5«alpha»,17«beta»)-

Other names:

5«alpha»-Androstane-3«alpha»,17«beta»-diol

Androstane-3«alpha»,17«beta»-diol

Etiocholane-3«alpha»,17«beta»-diol

Hombreol

3«alpha»,17«beta»-Dihydroxy-5«alpha»-androstane

5«alpha»-Androstane-3«alpha»,17«beta»-androstanediol

(3«alpha»,5«alpha»,17«beta»)-Androstane-3,17-diol

5-alpha-Androstan-3-alpha,17-beta-diol

NSC 9899

5A-Androstan-3A,17B-diol

Inchi:

InChI=1S/C19H32O2/c1-18-9-7-13(20)11-12(18)3-4-14-15-5-6-17(21)19(15,2)10-8-16(14)

InchiKey:

CBMYJHIOYJEBSB-ZOSDZXSZSA-N

Formula:

C19H32O2

SMILES:

CC12CCC3C(CCC4CC(O)CCC43C)C1CCC2O

Mol. weight [g/mol]:

292.46

CAS:

1852-53-5

## Physical Properties

Property code	Value	Unit	Source
gf	-23.86	kJ/mol	Joback Method
hf	-530.43	kJ/mol	Joback Method
hfus	26.87	kJ/mol	Joback Method
hvap	88.22	kJ/mol	Joback Method
log10ws	-4.66		Crippen Method
logp	3.751		Crippen Method
mcvol	246.870	ml/mol	McGowan Method
pc	1966.56	kPa	Joback Method
rinpol	2470.00		NIST Webbook
rinpol	2470.00		NIST Webbook
tb	848.59	K	Joback Method
tc	1062.44	K	Joback Method
tf	510.53	K	Joback Method
vc	0.917	m3/kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	907.30	J/mol×K	848.59	Joback Method
cpg	930.23	J/mol×K	884.23	Joback Method
cpg	953.16	J/mol×K	919.87	Joback Method
cpg	976.35	J/mol×K	955.52	Joback Method
cpg	1000.05	J/mol×K	991.16	Joback Method
cpg	1024.53	J/mol×K	1026.80	Joback Method
cpg	1050.06	J/mol×K	1062.44	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1852535&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1852535&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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