

# 3,3-Tetramethyleneglutaric anhydride

<b>Other names:</b>	«beta», «beta»-Tetramethyleneglutaric anhydride 1,1-Cyclopentanediacetic anhydride 8-Oxaspiro[4.5]decane-7,9-dione cyclopentane-1,1-diacetic anhydride 3,3-Tetramethyleneglutaric acid anhydride
<b>Inchi:</b>	InChI=1S/C9H12O3/c10-7-5-9(3-1-2-4-9)6-8(11)12-7/h1-6H2
<b>InchiKey:</b>	GFWLMILMVMCJDI-UHFFFAOYSA-N
<b>Formula:</b>	C9H12O3
<b>SMILES:</b>	O=C1CC2(CCCC2)CC(=O)O1
<b>Mol. weight [g/mol]:</b>	168.19
<b>CAS:</b>	5662-95-3

## Physical Properties

Property code	Value	Unit	Source
gf	-231.08	kJ/mol	Joback Method
hf	-479.95	kJ/mol	Joback Method
h <sub>fus</sub>	6.57	kJ/mol	Joback Method
h <sub>sub</sub>	96.40 ± 1.10	kJ/mol	NIST Webbook
h <sub>vap</sub>	48.30	kJ/mol	Joback Method
log <sub>10</sub> ws	-1.78		Crippen Method
logp	1.410		Crippen Method
m <sub>cvol</sub>	124.960	ml/mol	McGowan Method
pc	3990.60	kPa	Joback Method
tb	603.38	K	Joback Method
tc	870.54	K	Joback Method
tf	404.14	K	Joback Method
vc	0.456	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
c <sub>pg</sub>	337.14	J/mol×K	603.38	Joback Method
c <sub>pg</sub>	355.15	J/mol×K	647.91	Joback Method
c <sub>pg</sub>	372.02	J/mol×K	692.43	Joback Method

cpg	387.88	J/mol×K	736.96	Joback Method
cpg	402.88	J/mol×K	781.48	Joback Method
cpg	417.16	J/mol×K	826.01	Joback Method
cpg	430.86	J/mol×K	870.54	Joback Method
hfust	15.30	kJ/mol	338.40	NIST Webbook

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C5662953&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C5662953&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hsub:</b>	Enthalpy of sublimation at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/59-107-8/3-3-Tetramethyleneglutaric-anhydride.pdf>

Generated by Cheméo on 2024-04-17 02:17:53.953198085 +0000 UTC m=+15609522.873775408.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.