

Benzaldehyde, 2-hydroxy, 3,5-di(t-butyl)

Other names:	3,5-di-tert-butyl-2-hydroxybenzaldehyde
Inchi:	InChI=1S/C15H22O2/c1-14(2,3)11-7-10(9-16)13(17)12(8-11)15(4,5)6/h7-9,17H,1-6H3
InchiKey:	RRIQVLZDOZPJTH-UHFFFAOYSA-N
Formula:	C15H22O2
SMILES:	CC(C)(C)c1cc(C=O)c(O)c(C(C)(C)C)c1
Mol. weight [g/mol]:	234.33

Physical Properties

Property code	Value	Unit	Source
gf	-79.89	kJ/mol	Joback Method
hf	-419.73	kJ/mol	Joback Method
hfus	21.11	kJ/mol	Joback Method
hvap	69.73	kJ/mol	Joback Method
log10ws	-3.89		Crippen Method
logp	3.800		Crippen Method
mcvol	205.890	ml/mol	McGowan Method
pc	2276.24	kPa	Joback Method
rinpol	1800.00		NIST Webbook
tb	702.06	K	Joback Method
tc	930.12	K	Joback Method
tf	468.83	K	Joback Method
vc	0.729	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	663.43	J/molxK	930.12	Joback Method
cpg	597.31	J/molxK	740.07	Joback Method
cpg	612.03	J/molxK	778.08	Joback Method
cpg	625.87	J/molxK	816.09	Joback Method
cpg	638.96	J/molxK	854.10	Joback Method
cpg	651.43	J/molxK	892.11	Joback Method
cpg	581.56	J/molxK	702.06	Joback Method
dvisc	0.0000179	Paxs	663.19	Joback Method

dvisc	0.0000282	Paxs	624.32	Joback Method
dvisc	0.0000472	Paxs	585.44	Joback Method
dvisc	0.0000850	Paxs	546.57	Joback Method
dvisc	0.0001675	Paxs	507.70	Joback Method
dvisc	0.0000120	Paxs	702.06	Joback Method
dvisc	0.0003692	Paxs	468.83	Joback Method
psub	2.37e-04	kPa	300.17	Thermodynamic study on hydroxybenzaldehyde derivatives: 3- and 4-Hydroxybenzaldehyde isomers and 3,5-di-tert-butyl-2-hydroxybenzaldehyde
psub	2.34e-04	kPa	300.17	Thermodynamic study on hydroxybenzaldehyde derivatives: 3- and 4-Hydroxybenzaldehyde isomers and 3,5-di-tert-butyl-2-hydroxybenzaldehyde
psub	3.22e-04	kPa	302.18	Thermodynamic study on hydroxybenzaldehyde derivatives: 3- and 4-Hydroxybenzaldehyde isomers and 3,5-di-tert-butyl-2-hydroxybenzaldehyde
psub	3.04e-04	kPa	302.18	Thermodynamic study on hydroxybenzaldehyde derivatives: 3- and 4-Hydroxybenzaldehyde isomers and 3,5-di-tert-butyl-2-hydroxybenzaldehyde
psub	3.00e-04	kPa	302.18	Thermodynamic study on hydroxybenzaldehyde derivatives: 3- and 4-Hydroxybenzaldehyde isomers and 3,5-di-tert-butyl-2-hydroxybenzaldehyde
psub	4.08e-04	kPa	304.15	Thermodynamic study on hydroxybenzaldehyde derivatives: 3- and 4-Hydroxybenzaldehyde isomers and 3,5-di-tert-butyl-2-hydroxybenzaldehyde

psub	2.48e-04	kPa	300.17	Thermodynamic study on hydroxybenzaldehyde derivatives: 3- and 4-Hydroxybenzaldehyde isomers and 3,5-di-tert-butyl-2-hydroxybenzaldehyde
psub	3.78e-04	kPa	304.15	Thermodynamic study on hydroxybenzaldehyde derivatives: 3- and 4-Hydroxybenzaldehyde isomers and 3,5-di-tert-butyl-2-hydroxybenzaldehyde
psub	5.42e-04	kPa	306.37	Thermodynamic study on hydroxybenzaldehyde derivatives: 3- and 4-Hydroxybenzaldehyde isomers and 3,5-di-tert-butyl-2-hydroxybenzaldehyde
psub	5.14e-04	kPa	306.37	Thermodynamic study on hydroxybenzaldehyde derivatives: 3- and 4-Hydroxybenzaldehyde isomers and 3,5-di-tert-butyl-2-hydroxybenzaldehyde
psub	5.05e-04	kPa	306.37	Thermodynamic study on hydroxybenzaldehyde derivatives: 3- and 4-Hydroxybenzaldehyde isomers and 3,5-di-tert-butyl-2-hydroxybenzaldehyde
psub	6.73e-04	kPa	308.18	Thermodynamic study on hydroxybenzaldehyde derivatives: 3- and 4-Hydroxybenzaldehyde isomers and 3,5-di-tert-butyl-2-hydroxybenzaldehyde
psub	6.31e-04	kPa	308.18	Thermodynamic study on hydroxybenzaldehyde derivatives: 3- and 4-Hydroxybenzaldehyde isomers and 3,5-di-tert-butyl-2-hydroxybenzaldehyde

psub	6.20e-04	kPa	308.18	Thermodynamic study on hydroxybenzaldehyde derivatives: 3- and 4-Hydroxybenzaldehyde isomers and 3,5-di-tert-butyl-2-hydroxybenzaldehyde
psub	8.61e-04	kPa	310.17	Thermodynamic study on hydroxybenzaldehyde derivatives: 3- and 4-Hydroxybenzaldehyde isomers and 3,5-di-tert-butyl-2-hydroxybenzaldehyde
psub	8.12e-04	kPa	310.17	Thermodynamic study on hydroxybenzaldehyde derivatives: 3- and 4-Hydroxybenzaldehyde isomers and 3,5-di-tert-butyl-2-hydroxybenzaldehyde
psub	8.01e-04	kPa	310.17	Thermodynamic study on hydroxybenzaldehyde derivatives: 3- and 4-Hydroxybenzaldehyde isomers and 3,5-di-tert-butyl-2-hydroxybenzaldehyde
psub	1.10e-03	kPa	312.20	Thermodynamic study on hydroxybenzaldehyde derivatives: 3- and 4-Hydroxybenzaldehyde isomers and 3,5-di-tert-butyl-2-hydroxybenzaldehyde
psub	1.03e-03	kPa	312.20	Thermodynamic study on hydroxybenzaldehyde derivatives: 3- and 4-Hydroxybenzaldehyde isomers and 3,5-di-tert-butyl-2-hydroxybenzaldehyde
psub	1.01e-03	kPa	312.20	Thermodynamic study on hydroxybenzaldehyde derivatives: 3- and 4-Hydroxybenzaldehyde isomers and 3,5-di-tert-butyl-2-hydroxybenzaldehyde

psub	1.76e-04	kPa	298.16	Thermodynamic study on hydroxybenzaldehyde derivatives: 3- and 4-Hydroxybenzaldehyde isomers and 3,5-di-tert-butyl-2-hydroxybenzaldehyde
psub	1.80e-04	kPa	298.16	Thermodynamic study on hydroxybenzaldehyde derivatives: 3- and 4-Hydroxybenzaldehyde isomers and 3,5-di-tert-butyl-2-hydroxybenzaldehyde
psub	1.93e-04	kPa	298.16	Thermodynamic study on hydroxybenzaldehyde derivatives: 3- and 4-Hydroxybenzaldehyde isomers and 3,5-di-tert-butyl-2-hydroxybenzaldehyde
psub	1.40e-04	kPa	296.28	Thermodynamic study on hydroxybenzaldehyde derivatives: 3- and 4-Hydroxybenzaldehyde isomers and 3,5-di-tert-butyl-2-hydroxybenzaldehyde
psub	1.43e-04	kPa	296.28	Thermodynamic study on hydroxybenzaldehyde derivatives: 3- and 4-Hydroxybenzaldehyde isomers and 3,5-di-tert-butyl-2-hydroxybenzaldehyde
psub	1.51e-04	kPa	296.28	Thermodynamic study on hydroxybenzaldehyde derivatives: 3- and 4-Hydroxybenzaldehyde isomers and 3,5-di-tert-butyl-2-hydroxybenzaldehyde
psub	3.82e-04	kPa	304.15	Thermodynamic study on hydroxybenzaldehyde derivatives: 3- and 4-Hydroxybenzaldehyde isomers and 3,5-di-tert-butyl-2-hydroxybenzaldehyde

Sources

Thermodynamic study on hydroxybenzaldehyde derivatives: 3- and 4-hydroxybenzaldehyde isomers and 3,5-di-tert-butyl-2-hydroxybenzaldehyde	https://www.doi.org/10.1016/j.jct.2009.10.009
Joback Method	https://en.wikipedia.org/wiki/Joback_method
McGowan Method	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=R256818&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws

Legend

cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
psub:	Sublimation pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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