

Hexafluoromethanediamine

Other names:	Bis(difluoroamino)difluoromethane
Inchi:	InChI=1S/CF6N2/c2-1(3,8(4)5)9(6)7
InchiKey:	OLOSSWPNZKRPRU-UHFFFAOYSA-N
Formula:	CF6N2
SMILES:	FN(F)C(F)(F)N(F)F
Mol. weight [g/mol]:	154.01
CAS:	4394-93-8

Physical Properties

Property code	Value	Unit	Source
gf	-986.92	kJ/mol	Joback Method
hf	-455.30 ± 4.00	kJ/mol	NIST Webbook
hfus	15.45	kJ/mol	Joback Method
hvap	15.71	kJ/mol	Joback Method
log10ws	-2.06		Crippen Method
logp	1.679		Crippen Method
mcvol	55.530	ml/mol	McGowan Method
pc	4062.13	kPa	Joback Method
tb	239.55	K	Joback Method
tc	354.43	K	Joback Method
tf	171.93	K	Joback Method
vc	0.225	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	95.10	J/molxK	239.55	Joback Method
cpg	101.13	J/molxK	258.70	Joback Method
cpg	106.87	J/molxK	277.84	Joback Method
cpg	112.34	J/molxK	296.99	Joback Method
cpg	117.54	J/molxK	316.13	Joback Method
cpg	122.49	J/molxK	335.28	Joback Method
cpg	127.18	J/molxK	354.43	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C4394938&Units=SI

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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