

# Chloroacetic acid, 6-chlorohexyl ester

<b>Other names:</b>	6-chlorohexyl chloroacetate 1-Hexanol, 6-chloro, chloroacetate
<b>Inchi:</b>	InChI=1S/C8H14Cl2O2/c9-5-3-1-2-4-6-12-8(11)7-10/h1-7H2
<b>InchiKey:</b>	VNGMYNDYUJRSGC-UHFFFAOYSA-N
<b>Formula:</b>	C8H14Cl2O2
<b>SMILES:</b>	O=C(CCl)OCCCCCCI
<b>Mol. weight [g/mol]:</b>	213.10

## Physical Properties

Property code	Value	Unit	Source
gf	-241.30	kJ/mol	Joback Method
hf	-484.73	kJ/mol	Joback Method
hfus	27.66	kJ/mol	Joback Method
hvap	51.33	kJ/mol	Joback Method
log10ws	-2.34		Crippen Method
logp	2.568		Crippen Method
mcvol	155.500	ml/mol	McGowan Method
pc	2465.36	kPa	Joback Method
rinpol	1455.00		NIST Webbook
rinpol	1470.00		NIST Webbook
rinpol	1436.00		NIST Webbook
rinpol	1464.00		NIST Webbook
rinpol	1465.00		NIST Webbook
ripol	2211.00		NIST Webbook
ripol	2226.00		NIST Webbook
ripol	2211.00		NIST Webbook
ripol	2232.00		NIST Webbook
tb	533.59	K	Joback Method
tc	719.60	K	Joback Method
tf	311.92	K	Joback Method
vc	0.606	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	334.38	J/mol×K	533.59	Joback Method
cpg	345.75	J/mol×K	564.59	Joback Method
cpg	356.61	J/mol×K	595.59	Joback Method
cpg	366.99	J/mol×K	626.59	Joback Method
cpg	376.89	J/mol×K	657.60	Joback Method
cpg	386.31	J/mol×K	688.60	Joback Method
cpg	395.27	J/mol×K	719.60	Joback Method
dvisc	0.0027181	Paxs	311.92	Joback Method
dvisc	0.0014880	Paxs	348.87	Joback Method
dvisc	0.0009142	Paxs	385.81	Joback Method
dvisc	0.0006116	Paxs	422.75	Joback Method
dvisc	0.0004365	Paxs	459.70	Joback Method
dvisc	0.0003275	Paxs	496.64	Joback Method
dvisc	0.0002557	Paxs	533.59	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U330855&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U330855&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>ripola:</b>	Polar retention indices

**tb:** Normal Boiling Point Temperature  
**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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