

# Hydroperoxide, 1-methyl-1-phenylethyl

<b>Other names:</b>	1-Methyl-1-phenylethyl hydroperoxide 7-Cumyl hydroperoxide 7-Hydroperoxykumen CHP CHP-158 CHP-5 Cumeenhydroperoxyde Cumene hydroperoxide Cument hydroperoxide Cumenyl hydroperoxide Cumolhydroperoxid Cumyl hydroperoxide Hydroperoxide, «alpha», «alpha»-dimethylbenzyl Hydroperoxide, Â«alphaÂ», Â«alphaÂ»-dimethylbenzyl Hydroperoxyde de cumene Hydroperoxyde de cumyle Hyperiz Idroperossido di cumene Idroperossido di cumolo Isopropylbenzene hydroperoxide Kumenylhydroperoxid Percumyl H Rcra waste number U096 Trigonox K 80 Trigonox K-95 Trigonox R 239A UN 2116 «alpha», «alpha»-Dimethylbenzyl hydroperoxide «alpha»-Cumene hydroperoxide «alpha»-Cumyl hydroperoxide Â«alphaÂ», Â«alphaÂ»-Dimethylbenzyl hydroperoxide Â«alphaÂ»-Cumene hydroperoxide Â«alphaÂ»-Cumyl hydroperoxide
<b>Inchi:</b>	InChI=1S/C9H12O2/c1-9(2,11-10)8-6-4-3-5-7-8/h3-7,10H,1-2H3
<b>InchiKey:</b>	YQHLDYVWEZKEOX-UHFFFAOYSA-N
<b>Formula:</b>	C9H12O2
<b>SMILES:</b>	CC(C)(OO)c1ccccc1
<b>Mol. weight [g/mol]:</b>	152.19
<b>CAS:</b>	80-15-9

# Physical Properties

Property code	Value	Unit	Source
chl	-5108.20 ± 6.70	kJ/mol	NIST Webbook
gf	-101.67	kJ/mol	Joback Method
hf	-78.70 ± 6.70	kJ/mol	NIST Webbook
hfl	-149.00 ± 6.70	kJ/mol	NIST Webbook
hfs	-192.00	kJ/mol	NIST Webbook
hfus	10.97	kJ/mol	Joback Method
hvap	69.90	kJ/mol	NIST Webbook
hvap	69.90 ± 0.20	kJ/mol	NIST Webbook
hvap	70.30	kJ/mol	NIST Webbook
log10ws	-2.26		Crippen Method
logp	2.411		Crippen Method
mcvol	125.650	ml/mol	McGowan Method
pc	3642.13	kPa	Joback Method
tb	543.37	K	Joback Method
tc	748.48	K	Joback Method
tf	303.08	K	Joback Method
vc	0.458	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	293.53	J/molxK	543.37	Joback Method
cpg	305.74	J/molxK	577.56	Joback Method
cpg	317.17	J/molxK	611.74	Joback Method
cpg	327.86	J/molxK	645.93	Joback Method
cpg	337.84	J/molxK	680.11	Joback Method
cpg	347.17	J/molxK	714.30	Joback Method
cpg	355.85	J/molxK	748.48	Joback Method
dvisc	0.0105883	Paxs	303.08	Joback Method
dvisc	0.0030015	Paxs	343.13	Joback Method
dvisc	0.0011074	Paxs	383.18	Joback Method
dvisc	0.0004934	Paxs	423.23	Joback Method
dvisc	0.0002528	Paxs	463.27	Joback Method
dvisc	0.0001441	Paxs	503.32	Joback Method
dvisc	0.0000892	Paxs	543.37	Joback Method
hvapt	74.00	kJ/mol	368.50	NIST Webbook

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# Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	373.70	K	1.00	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.07891e+01
Coeff. B	-1.96802e+03
Coeff. C	-1.68886e+02
Temperature range (K), min.	356.29
Temperature range (K), max.	528.17

## Sources

McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C80159&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C80159&amp;Units=SI</a>
The Yaws Handbook of Vapor Pressure:	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

chl:	Standard liquid enthalpy of combustion
cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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