

# Phenol, 4-amino-

**Other names:**

- 1-Amino-4-hydroxybenzene
- 4-Amino-1-hydroxybenzene
- 4-Aminobenzenol
- 4-Aminophenol
- 4-Hydroxyaniline
- 4-hydroxy-1-aminobenzene
- 4-hydroxybenzenamine
- 4-hydroxyphenylamine
- Activol
- Azol
- BASF Ursol P Base
- Benzofur P
- C.I. 76550
- C.I. Oxidation Base 6
- C.I. Oxidation Base 6A
- Certinal
- Citol
- Durafur Brown RB
- Fouramine P
- Fourrine 84
- Fourrine P Base
- Furro P base
- Kodelon
- NSC 1545
- Nako Brown R
- PAP
- Para-aminophenol
- Paramidophenol
- Paranol
- Pelagol Grey P Base
- Pelagol P Base
- Phenol, p-amino-
- Renal AC
- Rodinal
- Takatol
- Tertral P Base
- UN 2512
- Unal
- Ursol P
- Ursol P Base

Zoba Brown P Base

p-Aminofenol

p-Aminophenol

p-Hydroxyaniline

p-Hydroxyphenylamine

**Inchi:** InChI=1S/C6H7NO/c7-5-1-3-6(8)4-2-5/h1-4,8H,7H2

**InchiKey:** PLIKAWJENQZMHA-UHFFFAOYSA-N

**Formula:** C6H7NO

**SMILES:** Nc1ccc(O)cc1

**Mol. weight [g/mol]:** 109.13

**CAS:** 123-30-8

## Physical Properties

Property code	Value	Unit	Source
chs	-3194.00	kJ/mol	NIST Webbook
chs	-3170.90 ± 0.50	kJ/mol	NIST Webbook
chs	-3167.40 ± 0.90	kJ/mol	NIST Webbook
gf	23.88	kJ/mol	Joback Method
hf	-81.50 ± 1.70	kJ/mol	NIST Webbook
hf	-90.50 ± 1.20	kJ/mol	NIST Webbook
hfs	-194.10	kJ/mol	NIST Webbook
hfs	-190.60 ± 0.90	kJ/mol	NIST Webbook
hfs	-168.00	kJ/mol	NIST Webbook
hfus	16.32	kJ/mol	Joback Method
hsub	103.63 ± 0.65	kJ/mol	NIST Webbook
hsub	109.10 ± 1.40	kJ/mol	NIST Webbook
hsub	109.10 ± 1.40	kJ/mol	NIST Webbook
hsub	103.60 ± 0.70	kJ/mol	NIST Webbook
hvap	54.88	kJ/mol	Joback Method
log10ws	-0.80		Aqueous Solubility Prediction Method
log10ws	-0.80		Estimated Solubility Method
logp	0.974		Crippen Method
mcvol	87.490	ml/mol	McGowan Method
pc	6328.92	kPa	Joback Method
rinpol	1314.00		NIST Webbook
rinpol	1314.00		NIST Webbook
rinpol	1265.00		NIST Webbook
rinpol	1265.00		NIST Webbook
rinpol	1314.00		NIST Webbook

rropol	1265.00		NIST Webbook
rropol	1314.00		NIST Webbook
tb	516.51	K	Joback Method
tc	762.01	K	Joback Method
tf	459.50 ± 0.50	K	NIST Webbook
tf	463.40 ± 1.00	K	NIST Webbook
tf	462.50 ± 1.00	K	NIST Webbook
tf	465.00 ± 2.00	K	NIST Webbook
tf	461.53	K	Aqueous Solubility Prediction Method
tf	457.00 ± 0.20	K	NIST Webbook
vc	0.259	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	191.19	J/mol×K	516.51	Joback Method
cpg	230.81	J/mol×K	721.09	Joback Method
cpg	224.18	J/mol×K	680.18	Joback Method
cpg	217.00	J/mol×K	639.26	Joback Method
cpg	209.16	J/mol×K	598.34	Joback Method
cpg	200.59	J/mol×K	557.43	Joback Method
cpg	236.99	J/mol×K	762.01	Joback Method
hfust	26.00	kJ/mol	462.50	NIST Webbook
hfust	31.20	kJ/mol	459.50	NIST Webbook
hfust	23.80	kJ/mol	455.20	NIST Webbook
hsubt	92.10	kJ/mol	416.50	NIST Webbook
hsubt	111.00	kJ/mol	441.00	NIST Webbook
hsubt	101.10 ± 0.70	kJ/mol	335.00	NIST Webbook

## Sources

Solubilities of p-Aminophenol in Sulfuric acid + Water from (286.15 to 329.15) K: and Viscosities of p-Aminophenol in Sulfuric Acid + Water from (293.15 to 343.15) K:

<https://www.doi.org/10.1021/je049552d>

Joback Method:

<https://www.doi.org/10.1021/je050443o>

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

Aqueous Solubility Prediction Method:

<http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDataset002.xlsx>

Estimated Solubility Method:

[http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl\\_file/ci034243xsi20040112\\_053635.txt](http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl_file/ci034243xsi20040112_053635.txt)

McGowan Method:

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C123308&Units=SI>

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hsub:</b>	Enthalpy of sublimation at standard conditions
<b>hsubt:</b>	Enthalpy of sublimation at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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