

# Benzenamine, 2-methoxy-4-nitro-

**Other names:** o-Anisidine, 4-nitro-  
p-Nitro-o-anisidine  
Amarthol fast red b base  
Azoamine pink o  
Azoene fast red b base  
Brentamine fast red B base  
c.i. Azoic diazo component 5  
c.i. 37125  
Dainichi fast red B base  
Daito red base B  
Devol red E  
Diabase red B  
Diazo fast red B  
Fast red B  
Fast red b base  
Fast red base b  
Fast red 5na base  
Hiltonil fast red b base  
Kako red b base  
Kayaku red b base  
Mitsui red b base  
Naphthanil red b base  
Naphtoelan red b base  
PNOA  
Red b base  
Red base ciba v  
Red base irga v  
Red base nb  
Sanyo fast red b base  
Shinnippon fast red b base  
Showa fast red b base  
Symulon red b base  
2-Amino-5-nitroanisole  
2-Methoxy-4-nitroaniline  
4-Amino-3-methoxynitrobenzene  
4-Nitro-o-anisidine  
4-Nitro-6-methoxyaniline  
5-Nitro-2-anisidine  
Anisole, 2-amino-5-nitro-  
Azoic diazo component 5, base

2-Amino-1-methoxy-5-nitrobenzene

2-Amino-5-nitroanisole

Aniline, 2-methoxy-4-nitro-

Naphthoelan red B base

Fast Red B-T Base

3-Nitro-6-aminoanisole

2-Methoxy-4-nitrobenzenamine

**Inchi:** InChI=1S/C7H8N2O3/c1-12-7-4-5(9(10)11)2-3-6(7)8/h2-4H,8H2,1H3

**InchiKey:** GVBHRNIWBGTNQA-UHFFFAOYSA-N

**Formula:** C7H8N2O3

**SMILES:** COc1cc([N+](=O)[O-])ccc1N

**Mol. weight [g/mol]:** 168.15

**CAS:** 97-52-9

## Physical Properties

Property code	Value	Unit	Source
chs	-3701.00 ± 0.96	kJ/mol	NIST Webbook
gf	98.21	kJ/mol	Joback Method
hf	-83.41	kJ/mol	Joback Method
hfs	-197.00	kJ/mol	NIST Webbook
hfus	24.89	kJ/mol	Joback Method
hvap	64.42	kJ/mol	Joback Method
log10ws	-1.88		Crippen Method
logp	1.186		Crippen Method
mcvol	119.000	ml/mol	McGowan Method
pc	4233.04	kPa	Joback Method
tb	642.99	K	Joback Method
tc	896.92	K	Joback Method
tf	469.21	K	Joback Method
vc	0.449	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	291.35	J/mol×K	642.99	Joback Method
cpg	301.87	J/mol×K	685.31	Joback Method
cpg	311.62	J/mol×K	727.63	Joback Method

cpg	320.60	J/mol×K	769.95	Joback Method
cpg	328.83	J/mol×K	812.28	Joback Method
cpg	336.32	J/mol×K	854.60	Joback Method
cpg	343.08	J/mol×K	896.92	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C97529&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C97529&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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