

# Benzoic acid, 2-hydroxy-, 2-propenyl ester

<b>Other names:</b>	Salicylic acid, allyl ester Allyl salicylate
<b>Inchi:</b>	InChI=1S/C10H10O3/c1-2-7-13-10(12)8-5-3-4-6-9(8)11/h2-6,11H,1,7H2
<b>InchiKey:</b>	YYKBLDCLPBWOOE-UHFFFAOYSA-N
<b>Formula:</b>	C10H10O3
<b>SMILES:</b>	<chem>C=CCOC(=O)c1ccccc1O</chem>
<b>Mol. weight [g/mol]:</b>	178.18
<b>CAS:</b>	10484-09-0

## Physical Properties

Property code	Value	Unit	Source
gf	-154.97	kJ/mol	Joback Method
hf	-309.88	kJ/mol	Joback Method
hfus	22.99	kJ/mol	Joback Method
hvap	61.63	kJ/mol	Joback Method
log10ws	-1.95		Crippen Method
logp	1.735		Crippen Method
mcvol	137.010	ml/mol	McGowan Method
pc	3857.88	kPa	Joback Method
rinpol	1339.00		NIST Webbook
ripol	1946.00		NIST Webbook
ripol	1946.00		NIST Webbook
tb	608.47	K	Joback Method
tc	836.63	K	Joback Method
tf	411.00	K	Joback Method
vc	0.459	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	329.63	J/molxK	608.47	Joback Method
cpg	340.98	J/molxK	646.50	Joback Method
cpg	351.54	J/molxK	684.52	Joback Method
cpg	361.38	J/molxK	722.55	Joback Method

cpg	370.57	J/mol×K	760.58	Joback Method
cpg	379.18	J/mol×K	798.60	Joback Method
cpg	387.29	J/mol×K	836.63	Joback Method
dvisc	0.0008638	Paxs	411.00	Joback Method
dvisc	0.0004122	Paxs	443.91	Joback Method
dvisc	0.0002178	Paxs	476.82	Joback Method
dvisc	0.0001250	Paxs	509.74	Joback Method
dvisc	0.0000767	Paxs	542.65	Joback Method
dvisc	0.0000498	Paxs	575.56	Joback Method
dvisc	0.0000339	Paxs	608.47	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C10484090&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C10484090&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/59-365-2/Benzoic-acid-2-hydroxy-2-propenyl-ester.pdf>

Generated by Cheméo on 2023-01-29 21:32:47.524027539 +0000 UTC m=+544306.926805192.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.