

# 4«beta»H,5«alpha»H-Eudesmane

|                      |  |
|----------------------|--|
| Inchi:               | InChI=1S/C15H28/c1-11(2)13-7-9-15(4)8-5-6-12(3)14(15)10-13/h11-14H,5-10H2,1-4H3/ |
| InchiKey:            | DYEQPYSFRWUNNV-URLJDCLXSA-N  |
| Formula:             | C15H28   |
| SMILES:              | CC(C)C1CCC2(C)CCCC(C)C2C1  |
| Mol. weight [g/mol]: | 208.38   |

## Physical Properties

| Property code | Value   | Unit    | Source         |
|---------------|---------|---------|----------------|
| gf            | 125.17  | kJ/mol  | Joback Method  |
| hf            | -262.69 | kJ/mol  | Joback Method  |
| hfus          | 14.80   | kJ/mol  | Joback Method  |
| hvap          | 47.34   | kJ/mol  | Joback Method  |
| log10ws       | -4.68   |         | Crippen Method |
| logp          | 4.885   |         | Crippen Method |
| mcvol         | 200.490 | ml/mol  | McGowan Method |
| pc            | 1872.41 | kPa     | Joback Method  |
| ripol         | 1404.00 |         | NIST Webbook   |
| ripol         | 1405.00 |         | NIST Webbook   |
| ripol         | 1578.00 |         | NIST Webbook   |
| ripol         | 1582.00 |         | NIST Webbook   |
| ripol         | 1578.00 |         | NIST Webbook   |
| tb            | 563.62  | K       | Joback Method  |
| tc            | 781.30  | K       | Joback Method  |
| tf            | 281.03  | K       | Joback Method  |
| vc            | 0.748   | m3/kmol | Joback Method  |

## Temperature Dependent Properties

| Property code | Value  | Unit    | Temperature [K] | Source        |
|---------------|--------|---------|-----------------|---------------|
| cpg           | 539.93 | J/molxK | 563.62          | Joback Method |
| cpg           | 566.03 | J/molxK | 599.90          | Joback Method |
| cpg           | 590.56 | J/molxK | 636.18          | Joback Method |
| cpg           | 613.67 | J/molxK | 672.46          | Joback Method |
| cpg           | 635.49 | J/molxK | 708.74          | Joback Method |

|     |        |         |        |               |
|-----|--------|---------|--------|---------------|
| cpg | 656.17 | J/mol×K | 745.02 | Joback Method |
| cpg | 675.83 | J/mol×K | 781.30 | Joback Method |

## Sources

|                        |   |
|------------------------|---|
| <b>Crippen Method:</b> | <a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>                                 |
| <b>Crippen Method:</b> | <a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>                         |
| <b>Joback Method:</b>  | <a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>                                     |
| <b>McGowan Method:</b> | <a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>                     |
| <b>NIST Webbook:</b>   | <a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R306413&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R306413&amp;Units=SI</a> |

## Legend

|                 |   |
|-----------------|---|
| <b>cpg:</b>     | Ideal gas heat capacity                         |
| <b>gf:</b>      | Standard Gibbs free energy of formation         |
| <b>hf:</b>      | Enthalpy of formation at standard conditions    |
| <b>hfus:</b>    | Enthalpy of fusion at standard conditions       |
| <b>hvac:</b>    | Enthalpy of vaporization at standard conditions |
| <b>log10ws:</b> | Log10 of Water solubility in mol/l              |
| <b>logp:</b>    | Octanol/Water partition coefficient             |
| <b>mcvol:</b>   | McGowan's characteristic volume                 |
| <b>pc:</b>      | Critical Pressure                               |
| <b>rinpol:</b>  | Non-polar retention indices                     |
| <b>ripol:</b>   | Polar retention indices                         |
| <b>tb:</b>      | Normal Boiling Point Temperature                |
| <b>tc:</b>      | Critical Temperature                            |
| <b>tf:</b>      | Normal melting (fusion) point                   |
| <b>vc:</b>      | Critical Volume                                 |

Latest version available from:

<https://www.chemeo.com/cid/59-386-9/4-beta-H-5-alpha-H-Eudesmane.pdf>

Generated by Cheméo on 2024-04-26 09:04:12.3269893 +0000 UTC m=+16411501.247566615.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.