

# Propane, 1,1,3,3-tetraethoxy-

<b>Other names:</b>	Malonaldehyde, bis(diethyl acetal) Malonaldehyde tetraethyl acetal Malonaldehyde tetraethyl diacetal Malondialdehyde tetraethyl acetal 1,1,3,3-Tetraethoxypropane Tetraethyl malondialdehyde acetal Malondialdehyde bis(diethyl acetal) Malonaldehyde diethyl acetal Tetraethoxypropane USAF KF-26 NSC 17068
<b>Inchi:</b>	InChI=1S/C11H24O4/c1-5-12-10(13-6-2)9-11(14-7-3)15-8-4/h10-11H,5-9H2,1-4H3
<b>InchiKey:</b>	KVJHGPAOUGYJX-UHFFFAOYSA-N
<b>Formula:</b>	C11H24O4
<b>SMILES:</b>	CCOC(CC(OCC)OCC)OCC
<b>Mol. weight [g/mol]:</b>	220.31
<b>CAS:</b>	122-31-6

## Physical Properties

Property code	Value	Unit	Source
gf	-383.14	kJ/mol	Joback Method
hf	-809.81	kJ/mol	Joback Method
hfus	21.95	kJ/mol	Joback Method
hvap	48.94	kJ/mol	Joback Method
log10ws	-1.99		Crippen Method
logp	2.175		Crippen Method
mvol	189.330	ml/mol	McGowan Method
pc	1848.33	kPa	Joback Method
ripol	1304.00		NIST Webbook
ripol	1347.00		NIST Webbook
tb	493.20	K	NIST Webbook
tc	707.42	K	Joback Method
tf	272.65	K	Joback Method
vc	0.712	m3/kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	471.94	J/molxK	539.88	Joback Method
cpg	487.68	J/molxK	567.80	Joback Method
cpg	502.93	J/molxK	595.73	Joback Method
cpg	517.70	J/molxK	623.65	Joback Method
cpg	531.95	J/molxK	651.57	Joback Method
cpg	545.69	J/molxK	679.50	Joback Method
cpg	558.90	J/molxK	707.42	Joback Method
dvisc	0.0029033	Paxs	272.65	Joback Method
dvisc	0.0010855	Paxs	317.19	Joback Method
dvisc	0.0005171	Paxs	361.73	Joback Method
dvisc	0.0002899	Paxs	406.26	Joback Method
dvisc	0.0001822	Paxs	450.80	Joback Method
dvisc	0.0001244	Paxs	495.34	Joback Method
dvisc	0.0000905	Paxs	539.88	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C122316&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C122316&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient

<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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