

N-(2-Ethoxyphenyl)-N-2,2,2-trifluoroacetyl-2,2,2-trifluoroethane

Inchi: InChI=1S/C12H9F6NO3/c1-2-22-8-6-4-3-5-7(8)19(9(20)11(13,14)15)10(21)12(16,17)18/1-12
InchiKey: WGJJMKHKRFBNIM-UHFFFAOYSA-N
Formula: C12H9F6NO3
SMILES: CCOc1ccccc1N(C(=O)C(F)(F)F)C(=O)C(F)(F)F
Mol. weight [g/mol]: 329.20

Physical Properties

Property code	Value	Unit	Source
gf	-1262.30	kJ/mol	Joback Method
hf	-1549.96	kJ/mol	Joback Method
hfus	31.55	kJ/mol	Joback Method
hvap	55.70	kJ/mol	Joback Method
log10ws	-3.63		Crippen Method
logp	3.070		Crippen Method
mcvol	185.790	ml/mol	McGowan Method
pc	2106.13	kPa	Joback Method
rinpol	1229.00		NIST Webbook
rinpol	1229.00		NIST Webbook
tb	637.38	K	Joback Method
tc	819.84	K	Joback Method
tf	426.88	K	Joback Method
vc	0.734	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	503.81	J/molxK	637.38	Joback Method
cpg	515.32	J/molxK	667.79	Joback Method
cpg	525.99	J/molxK	698.20	Joback Method
cpg	535.89	J/molxK	728.61	Joback Method
cpg	545.04	J/molxK	759.02	Joback Method
cpg	553.49	J/molxK	789.43	Joback Method
cpg	561.30	J/molxK	819.84	Joback Method

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U373207&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpola:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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